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Mercury pollution in Wuchuan mercury mining area, Guizhou, Southwestern China: The impacts from large scale and artisanal mercury mining

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To evaluate the environmental impacts from large scale mercury mining (LSMM) and artisanal mercury mining (AMM), total mercury (THg) and methyl mercury (MeHg) were determined in mine waste, ambient air, stream water and soil samples collected from Wuchuan mercury (Hg) mining area, Guizhou, Southwestern China. Mine wastes from both LSMM and AMM contained high THg concentrations, which are important Hg contamination sources to the local environment. Total gaseous mercury (TGM) concentrations in the ambient air near AMM furnaces were highly elevated, which indicated that AMM retorting is a major source of Hg emission. THg concentrations in the stream water varied from 43 to 2100 ng/L, where the elevated values were mainly found in the vicinity of AMM and mine waste heaps of LSMM. Surface soils were seriously contaminated with Hg, and land using types and organic matter played an important role in accumulation and transportation of Hg in soil. The results indicated heavy Hg contaminations in the study area, which were resulted from both LSMM and AMM. The areas impacted by LSMM were concentrated in the historical mining and smelting facilities, while Hg pollution resulted from AMM can be distributed anywhere in the Hg mining area.

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1. Introduction

Mercury (Hg) has been considered as a global pollutant [\(Lindqvist,](#page-6-0) [1991\)](#page-6-0) due to its long range transport in the atmosphere. Gaseous elemental mercury (GEM) (Hg⁰) is very stable with a residence time between 0.5 and 2 years [\(Schroeder and Munthe, 1998\)](#page-7-0). Its transport and deposition pose a serious environmental problem to the global environment, because the deposited Hg can be readily transformed into methyl mercury (MeHg) as a highly toxic form.

Mercury mining areas are considered as hot spots of Hg pollution, and the biogeochemical cycle of Hg has been widely studied worldwide in Hg mining areas: Almadén [\(Berzas Nevado et al.,](#page-6-0) [2003; Gray et al., 2004; Higueras et al., 2003; Millán et al., 2006](#page-6-0)) and Asturias [\(Loredo et al., 2005, 2006\)](#page-6-0) in Spain; Idrija ([Biester et al., 1999,](#page-6-0) [2000; Gosar et al., 1997\)](#page-6-0) in Slovenia; Amiata [\(Ferrara et al., 1998a\)](#page-6-0) in Italy; Nevada ([Bonzongo et al., 1996; Gray et al., 2002\)](#page-6-0), Alaska ([Bailey](#page-6-0) [et al., 2002; Gray et al., 2000](#page-6-0)) and California Coast Range [\(Domagalski,](#page-6-0) [2001; Domagalski et al., 2004; Ganguli et al., 2000; Kim et al., 2004](#page-6-0)) in U.S.; Palawan ([Gray et al., 2003\)](#page-6-0) in Philippines; Wanshan [\(Horvat et](#page-6-0) [al., 2003; Qiu et al., 2005; Zhang et al., 2010a, b\)](#page-6-0), Wuchuan ([Qiu et al.,](#page-6-0) [2006a; Wang et al., 2007a\)](#page-6-0), Tongren ([Li et al., 2008a](#page-6-0)) in Guizhou

Province and Xunyang ([Zhang et al., 2009\)](#page-7-0) in Shaanxi Province in China. The cinnabar ore roasting generally generates huge quantities of mine waste, which contain lots of fine-grained secondary Hg phases [\(Rytuba, 2003](#page-7-0)) and are dominant Hg sources to the local environment [\(Gray et al., 2003; Li et al., 2008a; Loredo et al., 2005\)](#page-6-0). Local air, surface water and soil can be seriously polluted by the emission and leaching of Hg from the mine waste calcines. The occupational exposures to Hg vapor ([Bellander et al., 1998; Gómez et al., 2007; Kobal et al., 2004; Li](#page-6-0) [et al., 2008b](#page-6-0)) as well as MeHg bio-accumulation in the food chain and related human health risk assessments [\(Feng et al., 2008; Li et al.,](#page-6-0) [2008c; Williams and Apostol, 2000\)](#page-6-0) in those Hg mining areas gained much concerns.

China has abundant Hg resources and Guizhou Province is the most important Hg production center ([Feng and Qiu, 2008](#page-6-0)). Wuchuan Hg mine is one of the largest Hg mines in Guizhou Province. More than 4070 tons of metal Hg had been produced during the period of large scale mercury mining (LSMM), which began in 1949 and ceased in 2003. Although Hg productions in China decreased temporarily in early 2000s with the closure of several largest mines, the internal demand has increased steadily, being estimated to be 700 tons in 2004 [\(NRDC, National Resource Defense Council, 2007\)](#page-6-0). The largest user of Hg is the manufacturing of polyvinyl chloride (PVC) plastic from coal ([NRDC, National Resource Defense Council, 2007; Hao et al.,](#page-6-0) [2005; Jian et al., 2009\)](#page-6-0). The increased demand stimulates the revival of artisanal mercury mining (AMM) in Hg mining areas in Guizhou

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Province. The annual Hg production of AMM was estimated to be 200 tons in whole China ([NRDC, National Resource Defense Council,](#page-6-0) [2007\)](#page-6-0) and 40 tons in Wuchuan mercury mining area (WMMA) ([Li](#page-6-0) [et al., 2009\)](#page-6-0).

A number of studies have been conducted in WMMA to investigate the influence by LSMM on the local Hg contamination [\(Qiu et al., 2006a; Qu, 2004; Wang et al., 2007a\)](#page-6-0). Concerning AMM, [Li](#page-6-0) [et al. \(2009\)](#page-6-0) estimated the annual Hg emission to the atmosphere to be 3.7–9.6 tonnes in 2004 and [Li et al. \(2008b\)](#page-6-0) evaluated the health aspects related to Hg vapor exposure for the artisanal mining workers.

Recent researches highlighted that rice cultivated in Hg contaminated areas can accumulate MeHg ([Qiu et al., 2008](#page-6-0)) and the main route of human exposure to MeHg is related to frequent rice consumption ([Feng et al., 2008](#page-6-0)). [Zhang et al. \(2010c\)](#page-7-0) demonstrated that rice preferentially bio-accumulate MeHg than inorganic Hg. [Meng et al. \(2010, 2011\)](#page-6-0) found that paddy soil is a potential source for MeHg in tissues of rice plant, where newly deposited Hg from AMM appear to be more susceptible to methylation than the bulk of old Hg in soil. This suggested the significant impact of AMM on MeHg accumulation in rice, which finally can pose a health risk to humans.

AMM is known for its low recovery efficiency, high environmental costs, and poor health and safety measures [\(Li et al., 2009\)](#page-6-0). It's periodically operated by the local peasants seasonally or according to the market price development. It can be widely distributed in the Hg mining area as a non-point pollution source to the local environment. This article is designed to investigate the extent and degree of Hg pollution to the local environment and to evaluate the contribution from LSMM and AMM in the study area.

2. Material and methods

2.1. Study area

Wuchuan is located in the northeast part of Guizhou Province and its detailed information has been given in previous studies [\(Li et al., 2008b;](#page-6-0) [Li et al., 2008c; Li et al., 2009; Qiu et al., 2006a\)](#page-6-0). The total area of Wuchuan County is approximately 2778 $km²$ and the Hg mining area is about 45 km². The main Hg mines and adits were located in Yinqiangou and Jixiuchang. The mineral processing plant was located in Laohugou, and the ore roasting ovens and distillation plants were distributed in Luoxi (Fig. 1). The mine waste calcines have been stockpiled into two collection dams located at Luoxi and Laohugou, respectively.

AMM in the study area was principally occurring around three villages of Yinqiangou, Luoxi and Taiba (Fig. 1). The supply of Hg ore for AMM in Yinqiangou and Luoxi was from the Hg mine by illegal exploitation; while in Taiba a local small Hg mine was the supplier. The procedures utilized in AMM have previously been described by [Li et al. \(2009\).](#page-6-0)

WMMA has an estimated population of 8500, of which agricultural population constitutes up to 85%. About 1000 people lived in the mine center, and most of them are laid-off workers in Wuchuan Hg mine. Other residents live in the surrounding villages, which are within 5 km of the mine center. Majority of the young people went out to do labor work in big cities. Some residents were involved in AMM activities in order to get rid of extreme poverty and landlessness. In addition to the laid-off workers in the mine center, the local inhabitants live on the foods they produce from their own land. They rarely eat fish, and rice and vegetables are the staple foods. The

Fig. 1. Location of sampling sites in the study area.

drinking water source for residents in mine center is from tap water, and for other residents is from local streams and ponds.

2.2. Sample collection

The environmental samples were collected in June 2005. Two control sites, Yanggang River and Taofu, were selected for the comparison with the polluted sites as shown in [Fig. 1.](#page-1-0)

Mine waste calcines were collected from Yinqiangou, Luoxi and Taiba as representative of AMM, and from Luoxi and Laohugou dam as representative of LSMM.

The total gaseous mercury (TGM) concentrations were measured in situ [\(Fig. 1\)](#page-1-0) with a portable atomic absorption spectrometer operating on the Zeeman principle (Lumex RA-915+, Russia). The detection limit of the instrument is 2 ng/m^3 . A single concentration data represents an average value of a time interval of 30 s and the measurements at each site were carried out at least 10 min.

Filtered and unfiltered water samples were collected in the study area ([Fig. 1\)](#page-1-0). Each sample site was selected at the point of the former large scale/artisanal Hg mining or at the fluvial system confluence with its tributaries. Water samples were filtered in situ through a 0.45 μm polyvinylidene fluoride filter (Millipore). Prior to sampling, the borosilicate glass bottles were pre-cleaned by heating for 1 h in a muffle furnace at 500 °C. Water samples were preserved by adding 0.4% (v/v) of ultra-pure HCl within 24 h. Collection, storage, and preservation techniques of water samples strictly followed the USEPA Method 1631 [\(USEPA, United States Environmental Protection](#page-7-0) [Agency, 1999](#page-7-0)).

Surface soil samples were collected at different villages to reveal spatial distribution of Hg pollution in the study area, which also include the two control sites. Moreover, a soil profile was collected at Jixiuchang to display vertical Hg distribution [\(Fig. 1\)](#page-1-0).

2.3. Analytical method

In the laboratory, the mine waste calcine and soil samples were air dried, ground, and sieved to minus 100 meshes particle sizes. For THg analysis, soil and mine waste samples were digested with fresh aqua regia (HCl + HNO₃, v/v 3:1) in water bath at 95 °C and determined by cold vapor atomic absorption spectrometry (CVAAS).

Soil pH was measured using an electrode with a solid:water ratio of 1:2.5. Organic matter (OM) contents in the soil samples were determined by a potassium dichromate volumetric method coupled with water heating technique [\(Lu, 2000](#page-6-0)).

Total Hg (unfiltered) and dissolved Hg (filtered) in stream water samples were determined by BrCl oxidation, $SnCl₂$ reduction, purge, gold trap and cold vapor atomic fluorescence spectrometry (CVAFS) detection ([USEPA, United States Environmental Protection Agency,](#page-7-0) [1999; Yan et al., 2003](#page-7-0)). Reactive Hg (RHg) was determined by addition of 20% SnCl₂ to unfiltered samples, followed by the purge, trap and CVAFS detection. The total MeHg (TMeHg) and dissolved MeHg (DMeHg) concentrations in the stream water samples were determined using distillation, $NABEt_4$ ethylation, and GC-CVAFS detection according to USEPA Method 1630 ([USEPA, United States](#page-7-0) [Environmental Protection Agency, 2001](#page-7-0)). The particulate Hg (PHg) and particulate MeHg (PMeHg) concentrations were obtained by subtracting dissolved fraction from THg and TMeHg, respectively.

2.4. Quality control

Quality control for THg and MeHg analysis was implemented by method blanks, blank spikes, matrix spikes, certified reference materials (GBW07405) as well as blind duplicates. The detection limits were 0.02 mg/kg for THg in mine waste and soil samples, whereas the detection limits for THg and MeHg in water sample were 0.02 ng/L and 0.009 ng/L, respectively. The average of THg

THg concentrations in the mine waste in the study area.

concentrations in the geological standard GBW07405 was determined to be 0.30 ± 0.01 mg/kg ($n=5$), which was comparable with the certified concentration of 0.29 ± 0.04 mg/kg. Recoveries on matrix spikes of MeHg in water samples were in the range of 93 to 106%. The relative standard deviation was $<$ 5% for THg in soil, mine waste and water duplicate samples and was $<10\%$ for MeHg in water duplicate samples.

3. Results and discussions

3.1. Mine wastes

THg concentrations in the mine waste calcines in the study area varied widely from 3.3 to 810 mg/kg (Table 1). THg concentrations in the mine waste calcines in Luoxi and Laohugou dam from LSMM and in Luoxi from AMM were generally very high $(>100 \text{ mg/kg})$. Highly elevated THg concentrations in the mine waste indicated that the retorting was not totally efficient [\(Rytuba, 2003](#page-7-0)). The results are similar to those found at Hg mines worldwide, e.g., Almadén, Spain and California and western Nevada, USA [\(Gray et al., 2002, 2004; Kim et al., 2004](#page-6-0)).

Generally, the mine waste calcines contain large quantities of secondary Hg phases, such as metacinnabar (a polymorph of HgS), corderoite, Hg chloride, Hg sulfates, Hg oxides, and elemental Hg ([Biester et al., 1999; Kim et al., 2000, 2004](#page-6-0)), which was the by-product during the distillation and condensation processes. These secondary Hg phases are more soluble than cinnabar and can cause a greater extent of Hg contamination to the surrounding environment. It may also result in a favorable condition for methylation of Hg in the surroundings. Previous studies indicated that mine waste calcines were important sources of MeHg to the downstream environment [\(Zhang et al., 2010b](#page-7-0)). The waste waters used for agricultural irrigation can provide abundant source of inorganic Hg and MeHg to the rice paddy field in the study area. The bio-accumulation of MeHg in rice and its consumption can cause a health threat to the local residents [\(Zhang et al., 2010a\)](#page-7-0).

Approximately 2 million tons of mine waste calcines had been dispersed into the adjacent ecosystems by LSMM. Meanwhile, 8500 tons of calcines were estimated to be discarded along the stream banks and flood plain by AMM ([Li et al., 2009\)](#page-6-0). Therefore, the huge quantities of calcines are potential threat to the local environment in the study area.

Table 2

Total gaseous mercury (TGM) concentrations in the study area.

Fig. 2. Total gaseous mercury (TGM) concentrations at different distances from AMM sites (Y-axis in log scale; a, Yinqiangou; b, Jixiuchang).

3.2. Ambient air

TGM concentrations in ambient air in the study area were shown in [Table 2.](#page-2-0) TGM concentrations were highly elevated at the sites of AMM. TGM concentrations near the smelting furnace in Yinqiangou, Jixiuchang, and Taiba were measured in μ g/m³ range with an average of 40, 30 and 1.2 μ g/m³, respectively. Air Hg concentrations near the smelting furnace were elevated by 3–4 orders of magnitude compared to that in the control sites. TGM concentrations near the smelting furnace in Yinqiangou and Jixiuchang exceeded the maximum allowable occupational standard in China (10 μ g/m³). Previous study indicated that the artisanal mining workers in the study area were seriously exposed to Hg vapor ([Li et al., 2008b](#page-6-0)).

Table 3

The comparison of TGM concentrations in Hg mining areas over the world.

200 m, 500 m and 800 m away from the smelting sites in the downwind direction, respectively. However, TGM concentration sharply decreased to a value of 47 ng/m³ at a distance of 100 m in the upwind direction. Hence, AMM activities represent the most important Hg emission source to the atmosphere in a regional scale. The annual atmospheric Hg emission from AMM was estimated to be 3.7–9.6 metric tons in 2004 in the study area [\(Li et al., 2009](#page-6-0)).

Highly elevated TGM concentrations were also found proximate to the abandoned facilities of LSMM. For instance, the average of TGM concentration near the Laohugou dam was 4500 ng/m³, whereas that close to the former Luoxi smelting plant was 230 ng/ $m³$. The results indicated that the mine waste and abandoned smelting facilities are also important Hg emission sources to the atmosphere. The annual Hg emissions from the mine waste heaps in whole Wanshan Hg mining area ranged from 0.96 to 3.0 kg ([Qiu, 2005](#page-6-0)).

TGM concentrations in ambient air at the two control sites were 18 and 28 ng/m³ respectively, which were comparable in magnitude to that of urban areas such as Guiyang City [\(Feng et al., 2004](#page-6-0)). Even though highly elevated compared to the global background values [\(Ebinghaus et al., 2002\)](#page-6-0), the results basically represented the regional background levels in the study area.

TGM concentrations in the study area are compared with those of other studies from Hg mines over the world as shown in Table 3. Our results were comparable with the observations reported during the Hg mine production period from Almadén Hg mine, Spain ([Ferrara et al., 1998b\)](#page-6-0), and with those reported for Wanshan and Wuchuan Hg mining impacted from AMM [\(Qiu, 2005; Wang et al., 2007a,b](#page-6-0)). [Bellander et al.](#page-6-0) [\(1998\)](#page-6-0) observed highly elevated TGM concentrations in the breathing air of the plant in Abbadia San Salvatore Hg mine, Italy with a maximum value of 234 μg/m³. However, in comparison with the typical air concentrations reported for abandoned Hg mines, the highly elevated data in this study indicated a significant impact of the AMM.

3.3. Stream water

Hg speciation and other parameters in the stream water in the study area are listed in [Table 4](#page-4-0). Low Hg levels were found in the control sites, due to a long distance to the Hg mining area and to a less impact from Hg mining activities. THg concentrations in the surface water at two control sites were 13 and 18 ng/L, respectively, which are comparable to the range (0.1–20 ng/L) of natural fluvial systems [\(Louis and Chamber](#page-6-0)[land, 1995](#page-6-0)).

THg concentrations in the stream water in the polluted area (with a range of 43– 2100 ng/L) were highly elevated compared to the control sites. Local maxima at sites 1, 2, and 7 possibly reflected the impact of atmospheric deposition and water leaching from the calcines both of which are related to the AMM. The Hg pollution at site 3 was derived from LSMM activities, since the influx flowed through the calcine pile in Laohugou dam. These results indicated that the mine wastes are significant Hg pollution sources to the local aquatic system.

The particulate Hg concentrations ranged from 28 to 2000 ng/L and constituted a large portion (65–96%) of the THg concentrations in the stream water. The reactive and dissolved Hg concentrations varied from 2.5 to 13 ng/L and from 8.7 to 95 ng/L, respectively. A significant correlation ($r = 0.999$, $n = 8$, $p < 0.001$; [Fig. 3](#page-4-0)a) was observed

AAS: atomic absorption spectrometry; LIDAR: light detection and ranging.

Type	No	pH	DO/ $mg \cdot L^{-1}$	DOC/ $mg \cdot L^{-1}$	THg/ $ng \cdot L^{-1}$	DHg/ $ng \cdot L^{-1}$	PHg/ $ng \cdot L^{-1}$	RHg/ $ng \cdot L^{-1}$	TMeHg/ $ng \cdot L^{-1}$	DMeHg/ $ng \cdot L^{-1}$	PMeHg/ $ng \cdot L^{-1}$	TMeHg/THg (%)
Polluted area	W1	8.5	7.54	3.04	2100	95	2000	13	0.38	0.28	0.10	0.02
	W ₂	8.6	7.56	1.19	960	40	920	7.6	0.45	0.44	0.017	0.05
	W ₃	8.5	7.10	0.66	280	24	260	10	0.59	0.41	0.18	0.21
	W ₄	8.6	7.23	2.29	510	77	430	6.0	0.41	0.35	0.063	0.08
	W ₅	8.3	7.53	2.51	180	19	160	4.1	0.31	0.26	0.057	0.17
	W ₆	8.4	6.42	8.36	43	15	28	6.5	0.43	0.31	0.12	1.00
	W7	8.6	7.64	2.13	670	32	640	11	1.1	1.0	0.10	0.16
	W ₈	8.6	8.42	0.40	190	8.7	180	2.5	0.43	0.29	0.14	0.23
Control site	W ₉	8.7	8.80	0.56	13	8.6	4.6	6.4	0.093	0.073	0.020	0.68
	W10	8.6	9.73	1.98	18	8.3	9.2	4.7	0.26	0.12	0.14	1.49

THg: total mercury; DHg: dissolved mercury; PHg (particulate mercury) = THg−DHg; RHg: reactive mercury; TMeHg: total methyl mercury; DMeHg: dissolved methyl mercury; PMeHg (particulate MeHg)= TMeHg−DMeHg.

between THg and particulate Hg concentration in the stream water. The results confirmed that Hg tends to be bound to particulate matters. Previous studies confirmed the increase of THg concentration with decreasing particle size in mine waste of Hg mine ([Kim et al., 2004](#page-6-0)). The suspended particles would present as a primarily transport pathway from mine waste to the aquatic system. With the deposition of suspended particles in downstream, therefore, the particulate Hg concentration decreased sharply as well.

Total and dissolved MeHg in water samples in the polluted area ranged from 0.31 to 1.1 ng/L and from 0.26 to 1.0 ng/L, respectively (Table 4). Considering the fraction of MeHg in the water samples, dissolved form is the primary species. On average, dissolved MeHg accounted for 80.1% of total MeHg in the water samples, with a range of 67.4–97.8%. A significant correlation ($r = 0.976$, $n = 8$, $p < 0.001$; Fig. 3b) was observed between total and dissolved MeHg concentrations in the stream water samples. Total

Fig. 3. The relationship between different Hg speciations in surface water in the study area.

MeHg concentrations in water samples in the polluted area were significantly higher than those in the control sites (0.093 and 0.26 ng/L). The highest MeHg concentration was found at Luoxi (site 7), where the former ore processing and retorting plant is located, and the long history of LSMM activities have resulted in serious Hg pollution to the local environment.

The proportion of THg present as MeHg in stream water was generally $<1\%$. However, a high ratio (1.49%) was found at the control site 10. However, no significant correlation between total MeHg and THg concentrations was observed in the water samples.

The range of THg concentrations in the stream waters in the study area was comparable to that obtained from other Hg mining areas (Table 5), while the MeHg concentrations were comparatively much lower. The results indicated the nonoptimum conditions of Hg methylation in the karstic environment. Alkalinity may significantly affect MeHg distribution in water body ([Ullrich et al., 2001](#page-7-0)). In the study area, the dominant mineral in Hg ore is cinnabar, which lacks other sulfide minerals. The host rocks are mainly carbonates, which contribute to the alkaline environment present in the calcines ([Zhang et al., 2004](#page-7-0)). The alkalinized stream water (pH 8.3–8.7) thus appears to inactivate bio-methylation processes in the environment.

Many studies suggested that dissolved organic carbon (DOC) may have mitigating effects on the production and/or bioaccumulation of MeHg in natural waters ([Ullrich et](#page-7-0) [al., 2001\)](#page-7-0). In this study, DOC contents in the water were comparatively low compared to that reported in relevant studies in North America and Europe [\(Gorski et al., 2003;](#page-6-0) [Watras et al., 1998\)](#page-6-0). High concentrations of inorganic carbon and low levels of complexing organic carbon may indirectly be responsible for the low levels of MeHg in the stream water.

3.4. Soils

THg concentrations in different types of arable soils are listed in [Table 6](#page-5-0). The data indicated that the surface soils were seriously contaminated with Hg (1.3–360 mg/kg) compared to that in the control sites (0.22–0.33 mg/kg). The latter represented the

Table 5

THg and MeHg concentrations in the stream water in Hg mines over the world/ng \cdot L⁻¹.

local geochemical baseline being in the range of 0.01–0.5 mg/kg present in worldwide uncontaminated soils [\(Senesi et al., 1999\)](#page-7-0).

Concerning co-located arable soils vegetated by various crops, generally the paddy field had higher THg concentrations than the vegetable and corn field. This may be due to the fact that the paddy fields were irrigated with highly Hg polluted stream water, and that the soils of paddy field contain abundant organic matters to complex and bind Hg [\(Biester et al., 2002; Lindqvist et al., 1991\)](#page-6-0). The water sources of vegetable and corn fields were mainly derived from rainfall and then atmospheric Hg deposition was the main Hg pollution source.

Highly elevated THg concentrations in the surface soil were mainly found in the areas seriously impacted by Hg mining activities, such as the vicinity of mine waste heaps, ore processing plant and retorting plant, and the sites of AMM. For instance, THg concentrations in the paddy field located at flood plains downstream of Laohugou dam and Luoxi were 360 and 220 mg/kg, respectively. These may have resulted from Hg rich particles transported from the upstream polluted areas during the flooding season. High THg concentrations were also found in the surface soil in Banchang and Ganxi, which may be impacted from the AMM.

The vertical distribution of THg in the soil profile from Jixiuchang exhibited a maximum near the surface (Fig. 4). With an increase of depth, the THg concentration decreased sharply and reached a stable value at the depth of 60–70 cm (12.8– 14.6 mg/kg, Fig. 4). THg concentrations at the bottom of the soil profiles were significantly higher than the local baseline (0.22–0.33 mg/kg), possibly explained by Hg mineralization in the bedrock. The trend of THg concentrations in the soil profile indicated that Hg contamination in top soil was primarily derived from the atmospheric deposition of Hg emitted from AMM.

No significant correlation was observed between soil pH and THg concentrations, however high THg concentrations were essentially related to high pH values (Fig. 5a). A significant correlation ($r = 0.53$, $p < 0.01$) was found between soil THg and OM concentrations (Fig. 5b). Organic matter has a strong capacity to complex and bind Hg ([Biester et al., 2002; Lindqvist et al., 1991](#page-6-0)). Positive correlations between soil organic carbon and THg concentrations have previously been reported [\(Hissler and](#page-6-0) [Probst, 2006; Tomiyasu et al., 2003](#page-6-0)). [Hissler and Probst \(2006\)](#page-6-0) concluded that most of atmospheric Hg deposition was trapped by OM contained in soil. The amount and quality of OM and its partition between aqueous and solid phases might play a major role in distribution and transportation of Hg through soil profiles ([Semu et al., 1987\)](#page-7-0). [Lin et al. \(2010\)](#page-6-0) reported that rice paddy soil contained a small portion, but significant amount of organically-bound Hg in Wanshan Hg mine. The accumulation of Hg in the organic-rich upper soil layers may be related to atmospheric Hg deposition.

Fig. 4. Total mercury (THg) distribution pattern in the soil profile at Jixiuchang in the study area.

4. Conclusions

This study revealed a significant Hg contamination in mine waste calcines, ambient air, stream water and soils in the study area, which was concluded as a combined impact of LSMM and AMM. The areas impacted by LSMM were essentially confined to the historical remains of Hg mining activities. However, AMM has a characteristic of nonpoint source pollution and the mercury pollution caused by AMM can be distributed widely over the Hg mining area. The mine waste calcines from AMM were important Hg contamination sources to the surface water and rice paddy fields. Therefore, the calcines must be appropriately disposed of and AMM should be strictly banned by the local government. More feasible remedial techniques should be

Fig. 5. The relationship between total mercury (THg) concentration and pH, organic matter (OM) in soil samples.

utilized to reduce Hg pollution and human exposure risks in the study area.

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