



Microstructure Evolution of Organic Matter and Clay Minerals in Shales with Increasing Thermal Maturity

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Abstract: As the two important components of shale, organic matter (OM) and clay minerals are usually thought to strongly influence the hydrocarbon generation, enrichment and exploitation. The evolution process of OM and clay minerals as well as their interrelationship over a wide range of thermal maturities are not completely clear. Taking Yanchang (T_{3y}), Longmaxi (S_1l) and Niutitang (C_{1n}) shales as examples, we have studied the microstructure characteristics of OM and clay minerals in shales with different thermal maturities. The effects of clay minerals and OM on pores were reinforced through sedimentation experiments. Using a combination of field emission scanning electron microscopy (FE-SEM) and low-pressure N_2 adsorption, we investigated the microstructure differences among the three shales. The results showed that both OM and clay minerals have strong effects on pores, and small mesopore (2–20 nm) is the dominant pore component for all three samples. However, the differences between the three samples are embodied in the distribution of pore size and the location. For the T_{3y} shale, clay minerals are loosely arranged and develop large amounts of pores, and fine OM grains often fill in intergranular minerals or fractures. Widespread OM pores distribute irregularly in S_1l shale, and most of the pores are elliptical and nondirectional. The C_{1n} shale is characterized by the preferred orientational OM-clay aggregates, and lots of pores in the composites are in the mesopore range, suggesting that over maturity lead to the collapse and compaction of pores under huge pressure of strata. The results of the current research imply that with increasing thermal maturity, OM pores are absent at low maturity (T_{3y}), are maximized at high maturity (S_1l) and are destroyed or compacted at over-mature stage (C_{1n}). Meanwhile, clay minerals have gone through mineral transformation and orientational evolution. The interaction of the two processes makes a significant difference to the microstructure evolution of OM and clay minerals in shale, and the findings provide scientific foundation in better understanding diagenetic evolution and hydrocarbon generation of shale.

Key words: organic matter, clay minerals, OM-clay composites, microstructure evolution, thermal maturity

Citation: Gu et al., 2020. Microstructure Evolution of Organic Matter and Clay Minerals in Shales with Increasing Thermal Maturity. Acta Geologica Sinica (English Edition), 94(2): 280–289. DOI: 10.1111/1755-6724.14285

1 Introduction

Shales are heterogeneous rocks with variations in structure and composition existing on many levels, from the nanoscale up to the macroscale (Curtis et al., 2012; Bernard and Horsfield, 2014). For most shales, organic matter (OM) and clay minerals, as the two important components of shale, play an important role in hydrocarbon generation, enrichment and exploitation. The microstructure can differ dramatically between shale plays and even within the same play (Curtis et al., 2010, 2012). Understanding how microstructure develops in OM and clay minerals is essential for predicting the storage and production capacity of hydrocarbons in shales (Kuila et al., 2014). Thermal maturity (R_o), to a certain extent, is a representative parameter to estimate the evolutionary history of shale. Therefore, the important thing in determining the potential of a shale to produce hydrocarbons is knowing the thermal maturity of the shale

(Curtis et al., 2012). Many previous studies have been performed on natural samples with different thermal maturities or artificially mature samples by pyrolysis experiments (Curtis et al., 2012; Chen and Xiao, 2014; Kuila et al., 2014; Romero-Sarmiento et al., 2014; Tang et al., 2015). For example, Curtis et al. (2012) investigated the organic porosity within Woodford Shale samples with vitrinite reflectance values ranging from 0.51% to 6.36%, and thought that maturity alone was not a reliable predictor of porosity in OM and other factors (such as OM composition) complicate this predication. Chen and Xiao (2014) analyzed nanopores characteristics of artificial shale samples across a maturation gradient by laboratory anhydrous pyrolysis and roughly subdivided evolution of the OM nanopores in shales into three stages: formation stage, development stage, and conversion or destruction stage. However, the evolution of OM and clay minerals as well as their interrelationship over a wide range of thermal maturities have not been thoroughly explored.

In the present study, three representative shales with different maturities in China, i.e., Yanchang (T_{3y}) shale,

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Longmaxi (S_1l) shale and Niutitang (C_{1n}) shale, were collected to investigate the microstructure characteristics of OM and clay minerals. The successful finding of industrial gas flow in T_{3y} has demonstrated the great potential of Ordos Basin (Guo et al., 2014; Dong et al., 2016). S_1l shale has been verified as a most commercially valuable shale gas reservoir in China. The S_1l shale, mostly distributed in the Sichuan Basin, has been exploited commercially in the Jiaoshiba shale gas field (Tan et al., 2015), with cumulative production from this formation exceeding $60 \times 10^8 \text{ m}^3$ in 2015 (Dong et al., 2016). The C_{1n} shale, widely developed in the Upper Yantze platform, has been considered a promising shale gas reservoir (Lu et al., 2015; Dong et al., 2016; Wang et al., 2016). The three shales (T_{3y} , S_1l , C_{1n}) represent low maturity (0.5–1.2%), high-over maturity (2.0–3.2%), and over maturity (3.0–5.0%), respectively (Er et al., 2013; Dong et al., 2016). Taking the three shales as examples, this study examined the microstructure evolution of OM and clay minerals in shale using low pressure N_2 adsorption and field emission-scanning electron microscopy (FE-SEM). Based on our results, we assessed the evolutionary pattern of OM and clay minerals with increasing thermal maturity, which would make a big difference in predicting the potential productivity of hydrocarbons in shales and provide scientific foundation for understanding the diagenetic evolution and hydrocarbon generation of shale.

2 Geological Settings

Previous studies have proved that the C_{1n} and S_1l shales were deposited under an epicontinental sea sedimentary environment (Liang et al., 2009). The lower strata of the C_{1n} and S_1l shales, as the main production layer of shale gas, are in a deep-water shelf environment (Lu et al., 2015; Yang et al., 2012; Zhang et al., 2012; Zhao et al., 2016), in which real advantages were provided for the formation of shale gas reservoirs by the high contents of silica and OM. As a typical continental shale in Ordos

Basin, the T_{3y} shale was deposited in silent deep lake, which was favorable for the development of organic-rich shale (Yuan et al., 2015).

The regional geology of the two basins is shown in Fig. 1. The Sichuan Basin is located at the transition zone between the Paleo-Pacific Tectonic Domain and the Tethys-Himalayan Tectonic Domain. The marine shales distributed in the Sichuan Basin have experienced substantial burial (high thermal maturation), significant uplift and erosion, and strong deformation from Mesozoic to Cenozoic (Liu et al., 2016). Around Sichuan Basin (Fig. 1a), Longmenshan and Dabashan thrust belts are distributed along with the northwest and northeast boundary of the basin, respectively, and Emei-Liangshan and Xiang-Qian-E fold belts are distributed at the southwest and southeast of the basin. In this study, the S_1l shale sample was collected from the southern end of Sichuan Basin, and the C_{1n} shale sample was obtained at the Xiang-Qian-E fold belt.

The Ordos Basin, located in northern-central China, is the second largest sedimentary basin in China, with vast oil and gas reserves (Dai et al., 2005; Duan et al., 2008; Yang et al., 2016b). The Ordos Basin is an intracratonic depression basin and one of the most tectonically stable basins in China (Dai et al., 2005; Guo et al., 2014). Six major structural units, including the Yimeng uplift in the north, the Weibei uplift in the south, the Tianhuan depression and Western edge thrust belt in the west, the Jinxi fold belt in the east, and the central Yishan slope (Fig. 1b) (Duan et al., 2008; Guo et al., 2014), compose the basin. The T_{3y} shale used in this paper was sampled from the southwest edge of the Yishan slope, which is the main oil and gas exploitation area in the basin (Dai et al., 2005; Guo et al., 2014).

3 Samples and Methods

3.1 Samples

Three samples from economically valuable source rocks were collected from a borehole to carry out this research.

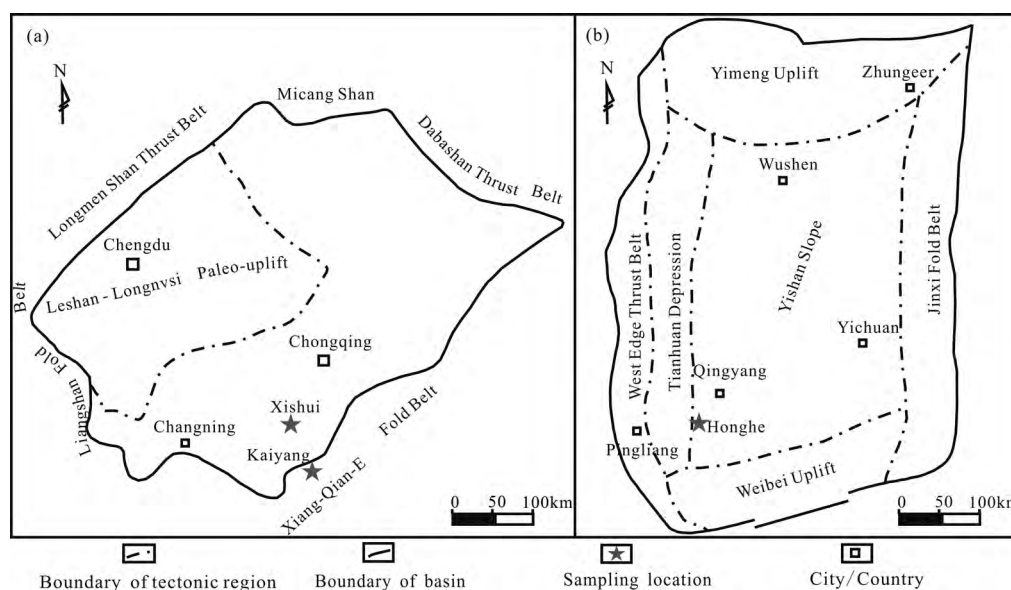


Fig. 1. The location of the sampling sites and the regional geology of Sichuan Basin (a) and Ordos Basin (b).

The first was collected from the continental shale of T₃y in Honghe-2 well, Ordos Basin, China. The second sample was mined at a depth of several meters from a shallow hole of S₁l in Xishui, Guizhou Province. The last sample was obtained from a drill of C₁n in Kaiyang, Guizhou Province. The location of the three sampling sites can be seen in Fig. 1.

3.2 Sedimentation experiments

To magnify the effects of OM and clay minerals on pore characteristics, sedimentation experiments were performed on the three samples. First, the three samples were crushed and ground to 200 mesh, and we called the powder primary samples. Then, we weighed a 5-g powdered sample into a beaker and added deionized water to soak the powder. All the samples were agitated and ultrasonicated to disperse the powder adequately. According to the Stokes sedimentation laws, the clay-sized was allowed to stand for 8 h, and the upper layer of the clay-sized (10 cm) was then extracted via siphon. After the extraction, an amount of deionized water equal to the amount above was added to the beaker, and the agitation and resting procedure was repeated twice. Next, the extracted clay-sized layer was centrifuged at 7000 r/min to remove the water, and then the solid was dried in an oven at 40°C (Ding et al., 2013). After being dried, the powder samples, which we called clay-sized samples, were ready for use in the next test. TOC content, mineralogical composition and low-pressure N₂ adsorption were performed on the primary samples and the clay-sized samples.

3.3 Organic geochemical analyses

The TOC of all samples was measured using a vario MACRO cube organic element analyzer following the Chinese Oil and Gas Industry Standard GB/T19145-2003. Before the measurement was taken, inorganic carbon in the samples was eliminated by the addition of diluted hydrochloric acid at a ratio of HCl: H₂O = 1:7 (V/V). The measurements were performed at State Key Laboratory of Environmental Geochemistry, Institute of Geochemistry, Chinese Academy of Sciences.

Previous researches on thermal maturity have indicated that laser Raman spectroscopy is a suitable method to calculate the reflectance (Kelemen and Fang, 2001; David, 2013; Chen et al., 2015). A Renishaw Invia Reflex Laser Raman Spectrometer at State Key Laboratory of Ore Deposit Geochemistry, Institute of Geochemistry, Chinese Academy of Sciences was used to measure the reflectance values of the three shales. In this work, the calculation of reflectance values (R) follows the equation: $R = 0.0537 d(G-D) - 11.21$, where G , D , and $d(G-D)$ denote peak positions of the graphitic carbon and disordered carbon and the interpeak intervals between G and D , respectively (Kelemen and Fang, 2001; Liu et al., 2012). For T₃y, the R represents vitrinite reflectance (R_o), while for S₁l and C₁n shales, due to the absence of vitrinite in the Lower Paleozoic marine shale, the bitumen reflectance (R_b) was measured to identify thermal maturity of samples. The relation $R_o = (R_b + 0.2443) / 1.0495$ is used to calculate R_o , according to the relationship between R_b and R_o

(Schoenherr et al., 2007).

3.4 Mineralogical composition

A PANalytical Empyrean X-ray diffractometer (XRD) was used to determine the mineralogical composition of the samples. The testing angle ranged from 5° to 80°. The results were analyzed quantitatively following the Chinese Oil and Gas Industry Standard (SY/T) 5163-2010. The tests were conducted at State Key Laboratory of Ore Deposit Geochemistry, Institute of Geochemistry, Chinese Academy of Sciences.

3.5 FE-SEM

Before the SEM observation, all samples were polished by a Technoorg SC1000 argon ion polisher operated at an 8-kV acceleration voltage to obtain a smooth surface. Then, the samples were sputter coated with gold. An FEI Scios FE-SEM at Center for Lunar and Planetary Sciences, Institute of Geochemistry, Chinese Academy of Sciences was used to characterize the distribution of nanoscale pores in shale samples.

3.6 Low-pressure N₂ adsorption

Low-pressure N₂ adsorption experiments were performed on a Quantachrome Autosorb-iQ2 automatic gas absorption analyzer at State Key Laboratory of Ore Deposit Geochemistry, Institute of Geochemistry, Chinese Academy of Sciences. To remove the free water and small amounts of volatile impurities, all samples were outgassed at 150 °C for 4 hours before adsorption measurements. The N₂ adsorption isotherms were obtained at 77 K (−196 °C), and the relative pressure (p/p_0) ranged from 10^{−6} to 0.99. The specific surface area (S_{BET}) was calculated using the multipoint Brunauer-Emmett-Teller (BET) method. The pore volume was acquired using the nonlocal density functional theory (NLDFT).

4 Results

4.1 Composition characteristics

The TOC, R_o and mineralogical composition of the three samples are shown in Table 1. TOC contents of these samples was 4.3% in T₃y, 4.% in S₁l, and 5.6% in C₁n, which suggests that all the three samples are organic-rich shale (>2.0 %). T₃y is immature source rock entering the oil window (1.1% R_o) whereas S₁l and C₁n are in the gas window (2.4–3.3% R_o). Various proportions of silicates (quartz, feldspar, and clay minerals), and sulfur-bearing minerals (pyrites) were evidenced from powder XRD. The overall mineralogical compositions were dominated by quartz, clay minerals (including illite, kaolinite and smectite), and feldspar. The presence of pyrite indicates that sulfate reduction took place in lacustrine (T₃y) or marine facies (S₁l and C₁n).

To evaluate nanoscale pore size distribution in clay minerals and OM deeply, sedimentation experiments were performed to extract clay minerals and OM. The TOC and mineralogical composition of the clay-sized samples were also tested (Table 2). The TOC values of the clay-sized samples were greater than those of the corresponding primary samples because of the low density of OM. For all

the three samples, the clay mineral contents of the clay-sized samples, principally for illite, were generally much greater than that of the primary samples. For the brittle minerals (quartz, feldspar, calcite, and dolomite), things just go opposite, i.e., brittle mineral contents of the clay-sized samples were generally less than that of the primary samples. The differences in mineralogical composition between the primary samples and the clay-sized samples should be caused by the smaller particle size and the lower density of the clay mineral content relative to the brittle minerals.

4.2 Adsorption characteristics

Low-pressure N_2 adsorption, as one of the most common techniques used to characterize nanoscale pore structures for shale, can provide a reliable assessment of the surface area, pore volume and pore size distribution (PSD). To develop a better understanding of N_2 adsorption behavior on OM and clay minerals in particular, both primary samples and the clay-sized samples were investigated by low-pressure N_2 adsorption and the N_2 adsorption isotherms are shown in Fig. 2. The isotherms of all samples can be classified as Type IIb according to a refined International Union of Pure and Applied Chemistry (IUPAC) grouping method (Rouquerol et al., 1994; Sing et al., 1985). The micropore (< 2 nm) filled by nitrogen molecules at extremely low relative pressure ($P/P_0 < 0.01$) results in the sharp increase in the adsorption volume. The monolayer-multilayer adsorption on mesopore (2–50 nm) walls occurs at intermediate P/P_0 . The absence of a plateau (as in mesoporous Type IV isotherms) and a steep slope in the P/P_0 range of 0.98–1.00 result from the existence of macropores (> 50 nm) (Rouquerol et al., 1994). All the T_{3y} and S_{1l} samples exhibit a hysteresis behavior, which shows the existence of mesopores, and the C_{1n} samples present an inconspicuous type H3 hysteresis loop. This isotherm type and hysteresis loop shape indicate that our samples contain aggregates of plate-like particles (e.g., clay minerals) and possess nonrigid slit-shaped pores (Rouquerol et al., 1994).

Like the primary samples, all the clay-sized samples present type IIb isotherms and an inconspicuous type H3 hysteresis loop, which indicates that there is little difference in the pore types between the primary samples and the clay-sized samples. However, there is a large difference in adsorption volume. At low $P/P_0 (< 0.01)$, the adsorption volumes of the S_{1l} shale primary samples are

larger than those of the clay-sized samples, while there is no clear difference for the T_{3y} and C_{1n} shale, which may be caused by the difference in the locations (clay or OM) of the micropores and mesopores between S_{1l} shale and C_{1n} shale. The adsorption volumes of the clay-sized samples are much larger than those of the primary samples for all three samples at high P/P_0 (0.98–1.00). The large increase of adsorption volumes at high P/P_0 can be attributed to the abundant macropores (either inherent or artificial pores created by particle packing) in clay-sized samples.

4.3 Microstructure characteristics

To further clear the microstructure characteristics of the three shale samples, FE-SEM was used to observe the morphology of clay minerals and OM directly. In the three samples, some common ground structure characteristics are illustrated by Figs. 3–5. First, OM and clay minerals are easily found in the three samples, indicating that both are indispensable in the generation, storage, and seepage of shale gas, though the great disparities of morphology exist among the three samples. Second, partial pyrite grains coexist with OM (Fig. 3a, c; Fig. 4a, c, f; Fig. 5a, b, d), which may be because both pyrite and OM prefer a reducing sedimentary environment. Last but not least, microfractures can be found between mineral grains and/or OM grains (Fig. 3a, c; Fig. 4b–d; Fig. 5a, e), which could have very important effects on the storage and permeable migration of shale gas (Hartman et al., 2011; Slatt and O'Brien, 2011; Ding et al., 2012; Dong et al., 2018).

Despite all the common characteristics described above, the three samples display a marked difference in pore distribution, morphological characteristics, and the relationship of OM and clay minerals. Figure 3 shows the microstructure characteristics of the T_{3y} shale, which is characterized by widely developed interparticle pores in clay minerals and microfractures. Most of the microfractures are filled with pyrite crystallites and organic matter (Fig. 3a, c), indicating the reducing sedimentary environment and that the microfractures are probably caused by locally abnormal pressure in organic matter evolution (Yang et al., 2016a). Many interparticle pores in clay minerals are linear in shape, whereas some pores have triangular shapes, which are defined by the lattice of randomly oriented clay mineral platelets (Fig. 3a–b, d–f). Most of the pores range between dozens and hundreds of nanometers in diameter. OM pores are not

Table 1 The burial depth, TOC, R_o and mineralogical composition of the three samples

Sample ID	burial depth (m)	R_o (%)	TOC (%)	quartz (%)	illite (%)	kaolinite (%)	smectite (%)	feldspar (%)	pyrite (%)	calcite (%)	dolomite (%)
T_{3y}	940.0	1.1	4.3	59.8	6.8	10.5	1.9	12.7	8.3	nd	nd
S_{1l}	5.0	2.4	4.8	37.2	17.3	6.2	nd	18.4	2.2	5.6	3.7
C_{1n}	964.5	3.3	5.6	36.4	24.1	12.1	nd	16.0	3.3	nd	8.1

Note: nd = not detected

Table 2 The TOC and mineralogical composition of the clay-sized samples

Sample ID	TOC (%)	quartz (%)	illite (%)	kaolinite (%)	smectite (%)	feldspar (%)	pyrite (%)	calcite (%)	dolomite (%)
T_{3y}	5.8	40.8	17.9	16.4	5.2	16.8	2.9	nd	nd
S_{1l}	6.4	37.8	37.1	6.5	nd	6.8	2.1	2.3	2.1
C_{1n}	7.7	16.9	55.3	13.5	nd	7.3	3.3	nd	2.4

Note: nd = not detected

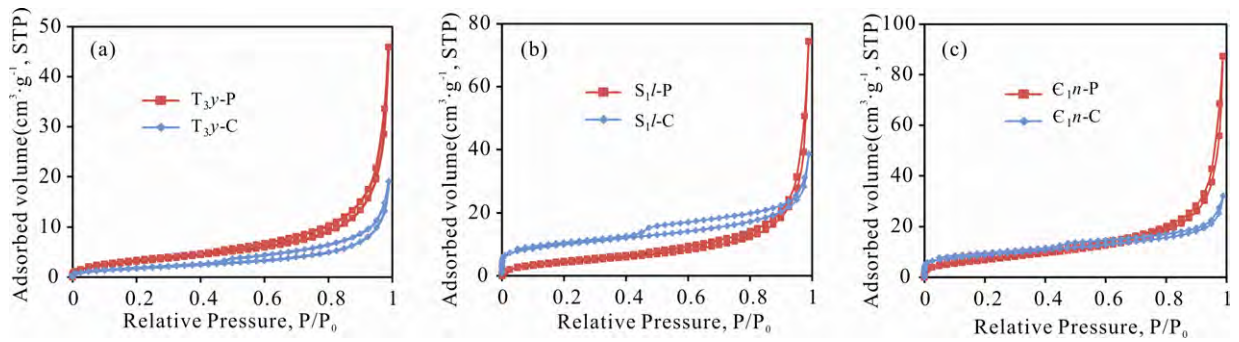


Fig. 2. N_2 adsorption-desorption isotherms of primary samples (P) and clay-sized samples (C). (a) presents the $T_{3\gamma}$ shale, (b) presents the S_{1I} shale, and (c) presents the C_{1n} shale.

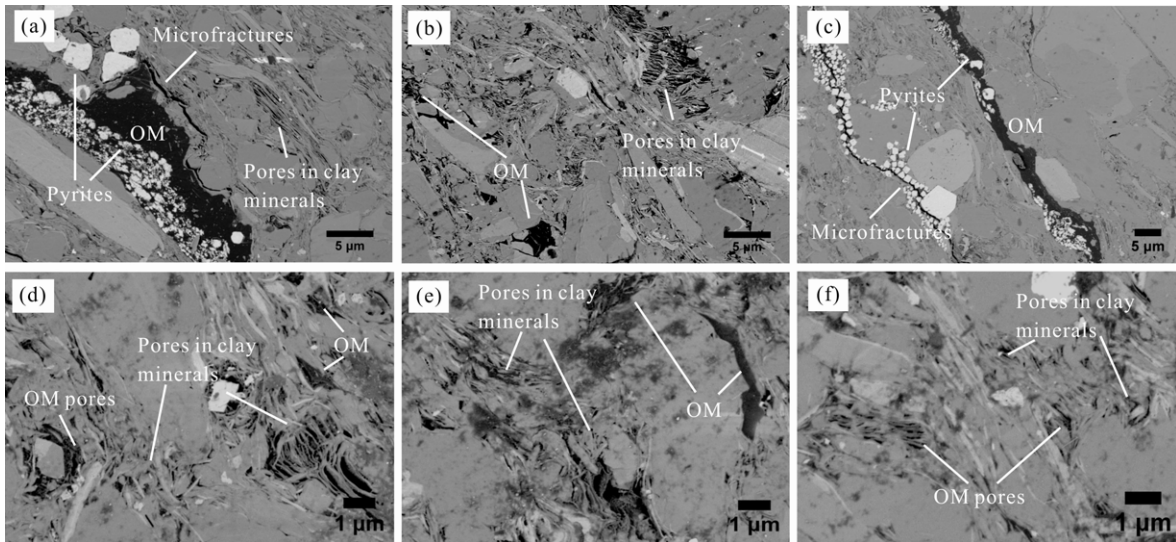


Fig. 3. FE-SEM images for the $T_{3\gamma}$ shale.

(a), (b) and (d)–(f) present the well-developed pores in clay minerals; (a) and (c) illustrate the coexistence of OM and pyrites, as well as microfractures; (a), (b) and (e) present the non-porous OM.

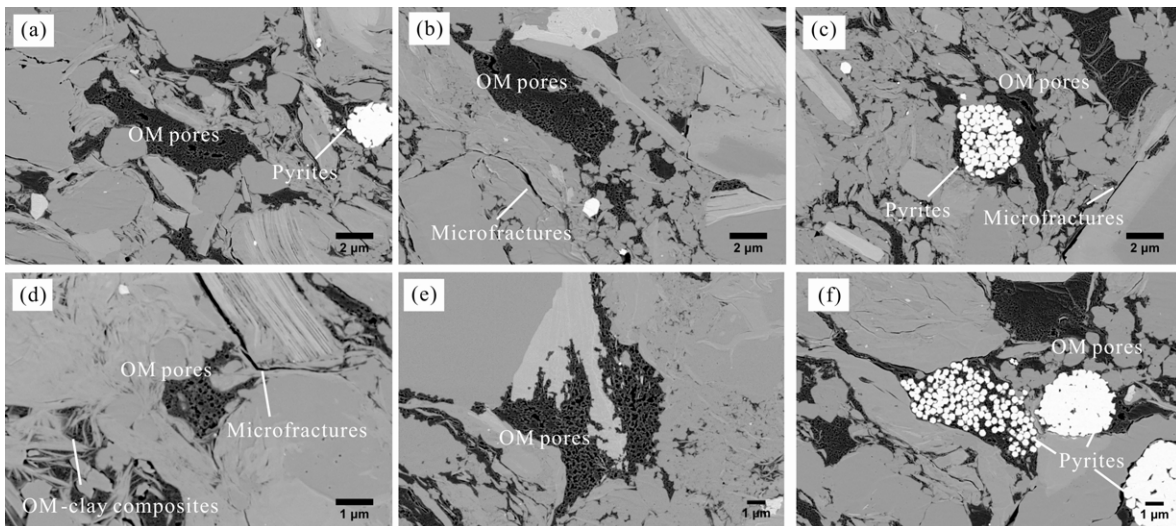


Fig. 4. FE-SEM images for the S_{1I} shale.

(a)–(f) present the well-developed OM pores; (a), (c) and (f) illustrate the coexistence of OM and pyrites; (b)–(d) present microfractures between minerals and/or OM grains.

developed and have a much smaller size than the pores in clay minerals (Fig. 3a–e). From Fig. 4, it is easily observed in S_{1I} shale that almost all of the organic grains

with diameters of several micrometers contain innumerable nanopores with various shapes and sizes. The OM pores are various in shape such as ellipsoid, elongated

bubble, irregular shape, but ellipsoid is the most common shape, and most of the OM pores lack a preferred orientation (Fig. 4). Several to a few hundred nanometers pores distribute widely in organic grains (Fig. 4), which is very similar to the dimension of OM pores investigated by Loucks et al. (2009, 2012). Fig. 5 presents few OM pores and widely developed interparticle pores in clay minerals and OM-clay composites for E_{1n} shale. The pores in OM-clay composites are most in several to tens of nanometers (Fig. 5c–e, f). In addition, the composites and OM grains have a preferred orientational arrangement (see the red dashed lines in Fig. 5d–f), which may be attributed to the release of internal overpressure in nanopores and compaction of pores under huge pressure of strata.

5 Discussion

5.1 Pore distribution characteristics

The PSD (NLDFT) results for the primary samples and clay-sized samples are shown in Fig. 6. The primary samples are dominated by mesopores for T_{3y} , and the plots of S_{1l} and E_{1n} present a clear bimodality with two peaks near 1 nm and 2–10 nm. Compared to primary samples, micropores are absent in the clay-sized samples of S_{1l} and E_{1n} , very likely because during sedimentation of the shale

samples, water molecules could occupy sorption sites in the micropores and block the pore throat by dispersed or dissolved clay mineral due to the strong water sensitivity of clay minerals (Chalmers and Bustin, 2007; Ju et al., 2014). The distribution of mesopore and macropore in the clay-sized samples are more obvious than in the primary samples, which indicates that mesopore and macropore are distributed mostly in OM or clay minerals.

In addition, the cumulative pore volume (CPV) curves for the primary samples and clay-sized samples always intersect (Fig. 6), indicating that the pore volumes of primary samples are larger than those of the clay-sized samples when the pore size is less than the intersection value, and when the pore size is greater than the intersection value, the pore volumes of primary samples are lower than those of the clay-sized samples. The intersections for the S_{1l} shale are larger than 20 nm but smaller than 20 nm for the T_{3y} and E_{1n} shales. To quantitatively evaluate PSD characteristics in more detail, according to pore size, the pores can be classified into four types: micropore (< 2 nm), small mesopore (2–20 nm), large mesopore (20–50 nm) and macropore (>50 nm).

The pore structure parameters of the three samples are presented in Table 3, including specific surface area

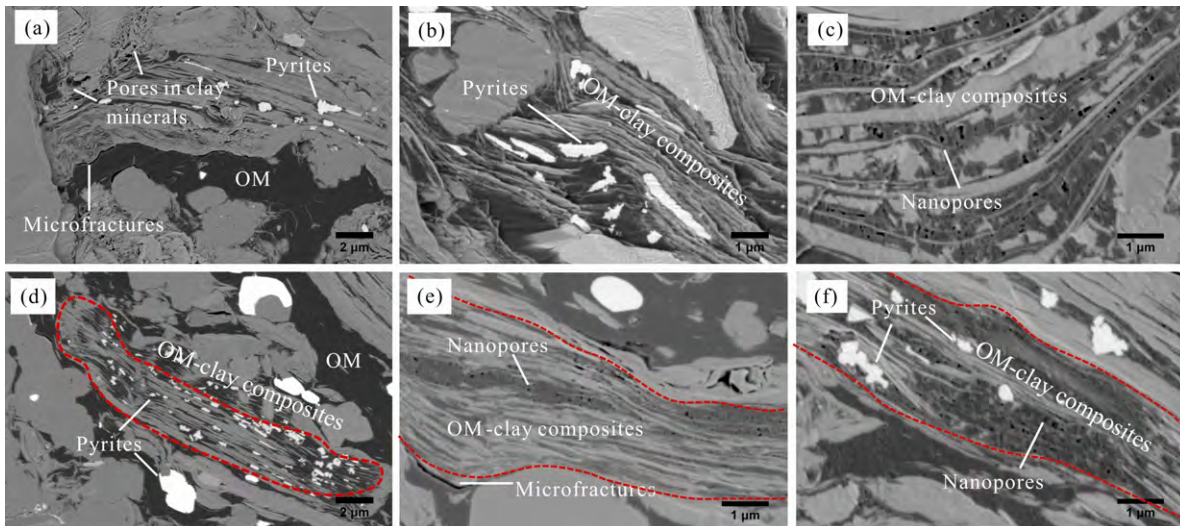


Fig. 5. FE-SEM images for the E_{1n} shale.

(a)–(f) indicate the exist of OM-clay composites; (c)–(f) present the well-developed pores in OM-clay composites; (a), (b) and (d) illustrate the coexistence of OM and pyrites; (a) and (e) present the microfractures between OM and clay minerals; (d)–(f) illustrate the preferred orientational arrangement of OM-clay composites (in or between the red dashed lines).

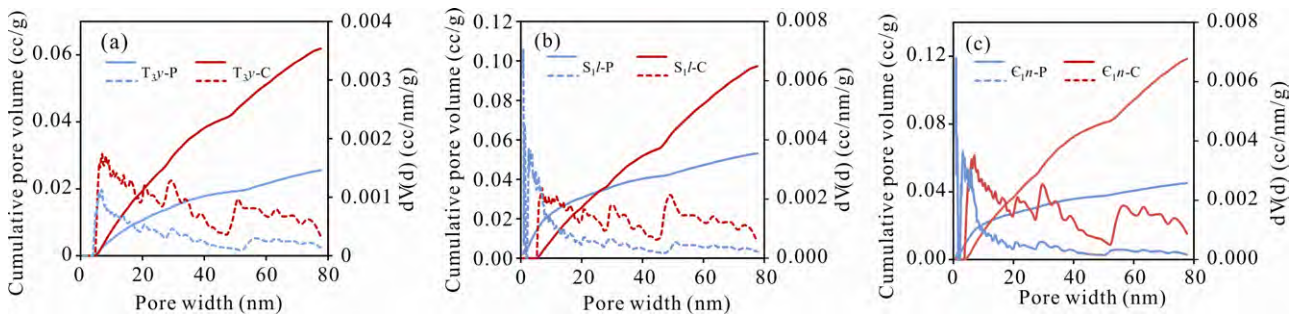


Fig. 6. Pore size distribution of primary samples (P) and clay-sized samples (C) for all samples.

(a) presents the T_{3y} shale, (b) presents the S_{1l} shale, and (c) presents the E_{1n} shale. Solid denotes CPV (cumulative pore volume) curves; Dashed denotes PSD (pore size distribution) curves.

(S_{BET}), average pore diameter (D_{ave}) and total pore volume (V_{tot}), micropore volume (V_{mic}), small mesopore volume (V_{2-20}), as well as large mesopore volume (V_{20-50}). Among the three samples, the sample T_{3Y} has the lowest S_{BET} and total pore volume (V_{tot}), as well as largest D_{ave} . The S_{BET} and V_{tot} of S_{1l} and C_{1n} shales are much larger than that of T_{3Y} , while the D_{ave} of S_{1l} and C_{1n} are much less than that of T_{3Y} . The higher TOC contents and R_o of S_{1l} and C_{1n} should be responsible for the trend. For all primary samples, D_{ave} (6.7 to 17.0 nm) fell within the small mesopore range, and V_{2-20} is larger than the other three types of pore (micropore, large mesopore and macropore) volume. Logically, small mesopore is clearly the dominant pore component in the three shale samples.

Table 4 presents the six pore structure parameters of the clay-sized samples. For T_{3Y} shale, because of the absence of OM pores (Fig. 3), no micropores can be detected in the sample. In S_{1l} and C_{1n} shales, V_{mic} of the clay-sized samples decreases to 0 cm^3/g , which may be because water molecules could occupy sorption sites in micropores during sedimentation (Chalmers and Bustin, 2007). Figure 7 illustrates the variation trends of the other five pore structure parameters (S_{BET} , V_{tot} , V_{2-20} , V_{20-50} , and D_{ave}) before and after extraction. Compared with primary samples, D_{ave} , V_{tot} and V_{20-50} of the clay-sized samples have presented apparent increase trends for all the three samples. The S_{BET} and V_{2-20} have exhibited different variation trends for the three shales (Fig. 7). Combining with FE-SEM observation, the differences between T_{3Y} , C_{1n} and S_{1l} suggest the differences of pore distribution regularities in clay minerals and OM. Previous studies have verified that micropore in shale reservoirs are mostly developed in OM, which can be concluded by the strong positive relationship between V_{mic} and OM content, while mesopore and macropore could develop in OM or clay minerals (Chalmers and Bustin, 2007; Chen and Xiao, 2014). Consequently, in contrast with primary samples, the clay-sized samples should have larger specific surface area and pore volume (including V_{tot} , V_{mic} , V_{2-20} and V_{20-50}). For the T_{3Y} shale, the increase in S_{BET} , V_{tot} , V_{2-20} and V_{20-50} should be interpreted as the development of pores in clay minerals (Fig. 3). However, things turned out differently for the S_{1l} shale. Figure 4 indicates the predominant control of OM on pores. The S_{BET} and V_{2-20} of the clay-sized sample are much less than those of the primary sample, which may be because of the decrease in V_{mic} and

that pore throat in OM could be blocked by dispersed or dissolved clay minerals due to the strong water sensitivity of clay minerals (Chalmers and Bustin, 2007; Ju et al., 2014). For the C_{1n} shale, the zeroth V_{mic} also leads to the small decrease in S_{BET} , but V_{2-20} of the clay-sized sample is larger than that of the primary sample because large amounts of small mesopores were found in OM-clay composites and clay minerals (Fig. 5), while the clay layers in OM-clay composites have a more stable laminated structure and are not easily affected by water molecules (Lu et al., 2013), resulting in the pores in OM-clay composites still being detected by nitrogen molecules after extraction. In addition, the increase in D_{ave} should be attributed to three aspects: the blockage of micropores or small mesopores (mainly for S_{1l}), the considerable development of large mesopore and macropore in OM and clay minerals (Figs. 3–5), as well as the macropores created manually during extraction.

5.2 The evolution of OM and clay minerals

The strong correlation between nanopore abundance in grains of OM and R_o has been proved in previous studies, suggesting that pore formation is the result of thermal maturation and conversion of OM (i.e., kerogen). This conclusion is supported by the abundance of nanopores in more thermally mature samples and the absence in OM grains from our lower thermal maturity samples (Loucks et al., 2007). However, when OM in shale evolves into over maturity, transformation and destruction of OM pores will occur. For example, Chen and Xiao (2014) analyzed the evolution process of nanoporosity in organic-rich shales during thermal maturation by conducting anhydrous pyrolysis experiments on shale samples and found that OM pores are enlarged after $R_o > 3.5\%$, with a decrease of V_{mic} , which can be attributed to the transformation of micropores to mesopores and/or macropores. For our three samples, OM pores are absent at low maturity (T_{3Y}) and reach the peak at high maturity (S_{1l}) and are destroyed or compacted at over-mature stage (C_{1n}) (Figs. 3–5), which implies that OM pore abundance is positive correlated with thermal maturity before the sample reaches high maturity, and then OM pores are destructed with further increasing thermal maturity. This evolutionary process is consistent with the study of Chen and Xiao (2014), i.e., in the formation and development stage, thermal maturation and conversion of OM could result in OM pores being more and more developed with increasing thermal maturity (from Fig. 3 to Fig. 4), while in the conversion or destruction stage, due to the carbonization and compaction of OM, a progressive decline happens to OM pores with further increase of maturity (from Fig. 4 to Fig. 5).

As one of the essential components of shale, clay minerals have significant action on the catalysis of organic hydrocarbon generation and physically protecting OM from degradation by external agents (Kennedy et al., 2014; Berthonneau et al., 2016). With increasing thermal maturity, the evolution of clay minerals includes mainly the dehydration and microstructural change (such as reduction of interplanar spacing) (Bray et al., 1998; Bala et al., 2000), as well as the transformation of mineral type (typically smectite to illite) (Eberl et al., 1993; Metwally

Table 3 Pore structure parameters of the three samples

Sample ID	S_{BET}^a (m^2/g)	D_{ave}^b (nm)	V_{tot}^c (cm^3/g)	V_{mic}^d (cm^3/g)	V_{2-20}^e (cm^3/g)	V_{20-50}^f (cm^3/g)
T_{3Y}	6.9	17.0	0.025	0	0.011	0.008
S_{1l}	35.4	6.7	0.053	0.005	0.026	0.012
C_{1n}	31.3	6.4	0.045	0.004	0.023	0.011

Note: ^a S_{BET} = specific surface area, ^b D_{ave} = average pore diameter, ^c V_{tot} = total pore volume, ^d V_{mic} = micropore volume, ^e V_{2-20} = small mesopore volume, ^f V_{20-50} = large mesopore volume.

Table 4 Pore structure parameters of clay-sized samples

Sample ID	S_{BET}^a (m^2/g)	D_{ave} (nm)	V_{total}^c (cm^3/g)	V_{mic}^d (cm^3/g)	V_{2-20}^e (cm^3/g)	V_{20-50}^f (cm^3/g)
T_{3Y}	12.4	23.0	0.062	0	0.019	0.024
S_{1l}	16.6	27.8	0.097	0	0.025	0.038
C_{1n}	26.7	20.2	0.118	0	0.037	0.044

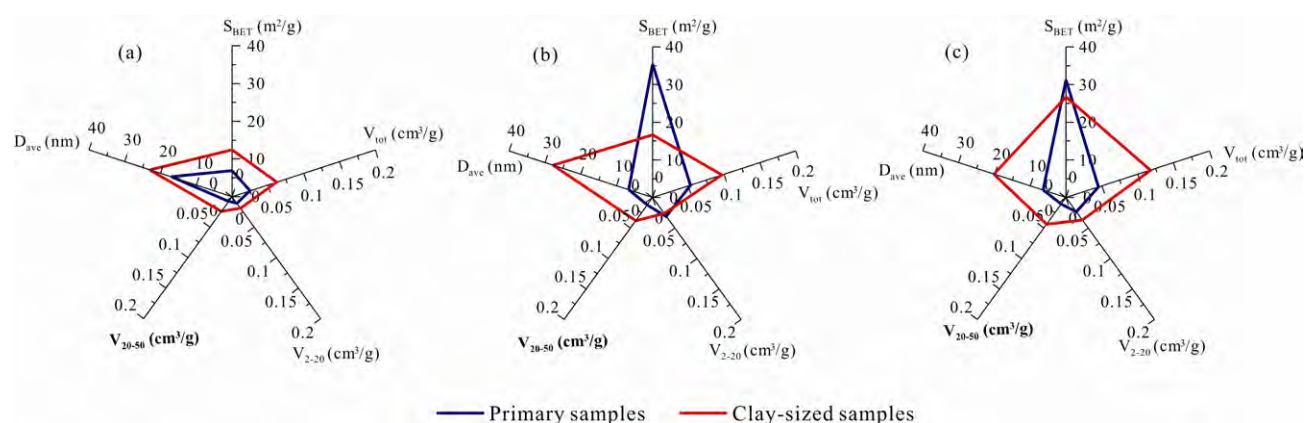


Fig. 7. The variation trends of five pore structure parameters (S_{BET} , V_{tot} , V_{2-20} , V_{20-50} , and D_{ave}) of the three samples before and after extraction.

(a) presents the T_{3y} shale, (b) presents the S_{1l} shale, and (c) presents the C_{1n} shale. Blue lines denote the primary samples, and red lines denote the clay-sized samples.

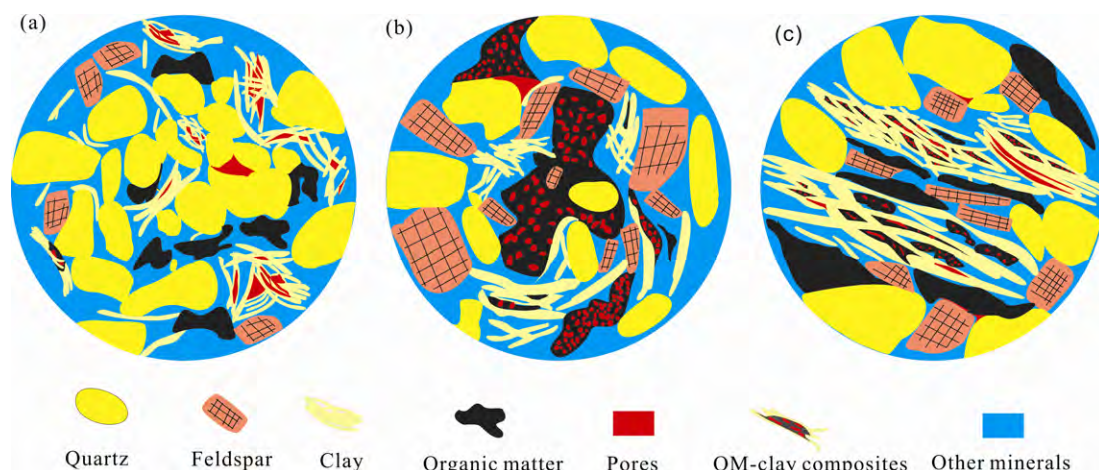


Fig. 8. Microstructure evolution of OM and clay minerals in shales with increasing thermal maturity.

(a) presents the low-mature shale (e.g., T_{3y} shale), (b) presents the high-mature shale (e.g., S_{1l} shale), and (c) presents the over-mature shale (e.g., C_{1n} shale).

and Chesnokov, 2012; Chen and Xiao, 2014; Berthonneau et al., 2016). For the three shale samples, there is some smectite in T_{3y} and none found in S_{1l} and C_{1n} (Table 1). Possibly, the over-high maturity of S_{1l} and C_{1n} shales has transformed smectite to illite. From the FE-SEM images, clay minerals are loosely arranged and are nondirectional in T_{3y} (Fig. 3). However, in S_{1l} and C_{1n} , accompanied by further maximum palaeoburial depths (6500 m for S_{1l} and 9000 m for C_{1n}) (Liu et al., 2016), clay minerals have encountered more serious compaction, so that clay particles integrate with each other more tightly. In addition, partial clay minerals integrated with OM and formed OM-clay composites (Figs. 4–5). OM-clay composites have been identified as a natural source material for hydrocarbon generation, and play an important role in the enrichment and preservation of organic matter (Lu et al., 2013; Kennedy et al., 2014; Zhu et al., 2016). With the increase of maturity, the microstructure of OM-clay composites also changes. First, the composites in S_{1l} are less developed than C_{1n} and not directional, but in C_{1n} , the composites seem to have a preferred orientational arrangement (Fig. 5d–f). With a slit-like pore geometry in OM-clay composites, it can be

argued that such pores would be more prone to be compacted due to increased effective pressure as gas is drawn out of the pores (Curtis et al., 2010). Second, the boundary lines between clay minerals and OM become even more blurred with the increase of maturity, which reflects that a closer relationship between clay minerals and OM often exists in more mature samples.

To sum up, the evolution of OM and clay minerals could interplay each other. As shown in Fig. 8, with increasing maturity (from a to b and then to c), OM pores would increase at first and then gradually disappear, and the pores in clay minerals tend to decrease in size and number, which may be attributed to the burial and thermal histories. The evolution of OM-clay composites has a directional trend, and in over-maturity, the pores developed in the composites are more numerous than the pores in OM or clay minerals, possibly because of the protection of clay layers on the OM. Consequently, thermal maturity, as one of the most important factors to affect microstructure of OM and clay minerals, has shown great significances in the research on occurrence, preservation mechanism, and hydrocarbon generation of OM.

6 Conclusions

Three representative shale samples (T_{3y} , S_{1l} , C_{1n}) with different maturities in China were investigated to analyze the microstructure evolution of OM and clay minerals. The clay-sized samples extracted through sedimentation experiments were measured to magnify the effects of OM and clay minerals on pores. Preliminary conclusions are as follows:

(1) Both OM and clay minerals have strong effects on pore characteristics with increasing thermal maturity. For T_{3y} shale, clay minerals arrange loosely and develop large numbers of pores, ranging from several to hundreds of nanometers. Elliptical OM pores widespread in the S_{1l} shale, and are mostly in the range of mesopore to macropore. The well-developed OM-clay aggregates in C_{1n} shale seem to have a preferred orientation, and lots of mesopores are found in the composites.

(2) Small mesopores (2–20 nm), which developed mostly in clay minerals for T_{3y} , in OM for S_{1l} , and in OM-clay minerals for C_{1n} shale, are the dominant pore component among the three samples. The distribution differences of pores imply the evolutionary processes of OM and clay minerals in shale. With the increase of maturity, OM pores are absent at low maturity (T_{3y}) and reach the peak at high maturity (S_{1l}) and are destroyed or compacted at over-mature stage (C_{1n}), and clay particles, which have gone through mineral transformation and orientational evolution, integrate with each other more tightly. The interactions of the two processes have a great significance for the microstructure evolution of OM and clay minerals in shale.

(3) The evolution of OM and clay minerals could interplay each other, and makes the two components get closer, which is embodied in the boundaries between OM and clay minerals becoming more blurred with the sample maturity increasing. The changes of their relationship with different thermal maturity should be of great significance in the research on occurrence, preservation mechanism, and hydrocarbon generation of OM.

Acknowledgments

This work is supported by the Chinese Academy of Sciences (“Hundred Talents Program”) and the National Natural Science Foundation of China (41802143).

Manuscript received Oct. 30, 2018

accepted Apr. 17, 2019

associate EIC: HAO Ziguo

edited by FEI Hongcai

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