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Helium and argon isotope geochemistry of the Tibetan Qulong porphyry Cu-Mo deposit, China

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Abstract The Oulong porphyry Cu–Mo deposit, generated in the Miocene post-collisional extension environment of the Gangdese Copper (Molybdenum) Metallogenic Belt, is one of the largest porphyry Cu deposits in China. This study reports the noble gas isotopic compositions of volatiles released from fluid inclusion reserved in pyrite from the Oulong deposit. ³He/⁴He and ⁴⁰Ar/³⁶Ar ratios range from 0.54 to 1.015 Ra and 300-359, respectively. Concentrations of ⁴He and ⁴⁰Ar range from 1.77 to $2.62 \times 10^{-8} \text{ cm}^3 \text{ STP}$ and $1.7-34 \times 10^{-8} \text{ cm}^3 \text{ STP}$, respectively. The isotopic composition of noble gases indicates that the ore-forming fluids of the Oulong Cu-Mo deposit were a mixture of fluid containing mantle component, which is exsolved from the porphyry magma, and crustal fluid characterized by atmospheric Ar and crustal radiogenic He. The δ^{34} S values of pyrite and molybdenite range from -0.52% to 0.31%, with an average of -0.12%, indicating a magmatic origin. More mantle components were involved in the Cu-Mo deposit than in the Mo-Cu deposit in the Qulong-Jiama ore-district.

Keywords He and Ar isotopes · Sulfur isotope · Qulong Cu–Mo deposit · Ore-forming fluid · Tibet · China

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1 Introduction

The Qulong porphyry Cu-Mo deposit (PCD), located in the east part of the Gangdese copper Metallogenic Belt, about 50 km east of Lhasa city, is one of the largest porphyry Cu deposits in China, with reserves of 10.6 Mt Cu@0.5% and 0.5 Mt Mo@0.03% (Xiao et al. 2012). Many studies have been done on the geology (Yang 2008; Yang et al. 2009; Qin et al. 2014), petrology (Hu et al. 2015; Mo et al. 2006; Yang et al. 2011), mineralization (Meng et al. 2006), geochronology (Meng et al. 2003; Wang et al. 2006; Li et al. 2017a, b), magmatic-hydrothermal evolution (Li et al. 2017a; Yang et al. 2005, 2009), and thermal history (Zhao et al. 2016; Zhou et al. 2019) of the Qulong deposit. Approximately the same S and Pb isotope compositions of sulfide and ore-related porphyry show that the ore-forming materials are mainly from the porphyry magma (Ou et al. 2002, 2007; Meng et al. 2006). However, the contribution of the mantle is not clear.

Among the noble gases, He and Ar are the most efficient for tracing the sources of ore-forming fluids because of the significant difference of isotopic compositions between end members. ³He/⁴He ratio of the upper mantle is 7–9 Ra (Graham, 2002, where Ra is the atmospheric ${}^{3}\text{He}/{}^{4}\text{He}$ ratio and 1 Ra = 1.39×10^{-6}), while the ³He/⁴He ratio of crustal rocks is 0.01–0.05 Ra (O'Nions and Oxburgh 1983), ⁴⁰Ar/³⁶Ar ratio of air is 295.5, while mantle-derived Ar is dominated by radiogenic ⁴⁰Ar, with ⁴⁰Ar/ 36 Ar > 40,000, and crustal rocks have ${}^{40}\text{Ar}/{}^{36}\text{Ar} \ge 45,000$ due to radiogenic and nucleogenic Ar produced in the crust. Therefore, ⁴⁰Ar/³⁶Ar ratios are sensitive to trace the addition of meteoric fluid from crust or mantle Ar, and the ${}^{3}\text{He}/{}^{4}\text{He}$ ratios can distinguish the involvement of mantle He (Stuart et al. 1995; Hu et al. 1998, 1999, 2004, 2009, 2012). Thus, He and Ar isotopes can provide a unique insight into the

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origin and evolution of ore-forming fluids (Burnard et al. 1999; Burnard and Polya 2004; Davidheiser-Kroll et al. 2014; Hu et al. 2012; Kendrick et al. 2001; Stuart et al. 1995; Wu et al. 2018).

In this study, He and Ar isotopes of fluid inclusions in pyrites and S isotopic compositions of pyrite and molybdenite were obtained to constrain the origin of the oreforming fluids of the Qulong PCD.

2 Geological background

The Qulong PCD is situated in the southern part of the Lhasa Terrane, which is bordered by the Bangong-Nujiang suture (BNS) in the north and the Indus-Yarlung suture (IYS) in the south (Fig. 1), and is composed mainly of the Paleozoic to Mesozoic marine clastic sedimentary rocks and volcanic rocks, and the Paleozoic to Cenozoic magmatic rocks, with the Precambrian basement known as the Nyaingêntanglha Group (Yang et al. 2009; Zhang et al. 2014). The Lhasa terrane and is divided into the northern, central, and southern Lhasa subterranes, separated by the Shiquan River-NamTso Mélange and the Luobadui-Milashan faults (Zhang et al. 2014). According to the zircon $\varepsilon_{\rm Hf}$ values and relevant Hf model ages, the south and north Lhasa subterranes are dominated by juvenile crustal blocks while the central Lhasa subterrane is characterized with ancient and locally reworked crust (Hou et al. 2015).

The southern Lhasa subterrane mainly consists of Gangdese volcanic-magmatic arc, dominated by the extensively distributed Early Cenozoic Linzizong volcanic succession and the Cretaceous-Tertiary granitic intrusions (Pan et al. 2004, Pan 2006; Zhu et al. 2008, 2013). The India-Asia collision initiated along the Indus-Yarlung Zangbo suture at around 65-55 Ma (Rowley 1996; Mo et al. 2003; Wang et al. 2003; Ding et al. 2005; Dong et al. 2006; Pan 2006; Zhu et al. 2011, 2013), followed later by the subduction of India continental plate at around 35 Ma (Ali and Aitchison 2008; Wang et al. 2014). The subsequent break off of the denser India plate beneath the Asia plate at 25-15 Ma (Aitchison et al. 2007; Van Hinsbergen et al. 2012), which triggered the generation of the Oligo-Miocene magmatic rocks associated with porphyry Cu-Mo deposits along the of the Gangdese porphyry copper belt (Hou et al. 2009; Van Hinsbergen et al. 2012; Wang et al. 2014; Li et al. 2017a, b), of which the Qulong deposit is the largest (Zheng et al. 2004; Qin et al. 2014). The porphyry magmatism culminated at 16 ± 1 Ma, ranging from 26 to 13 Ma, and the molybdenite Re-Os ages of the mineralization are from 17 to 14 Ma (Hou et al. 2015 and the references therein).

The central Lhasa subterrane is composed of many Miocene porphyry (20–16 Ma) Mo–Cu deposits, such as the Bangpu porphyry Mo–Cu deposit. Most of those deposits are situated along the southern margin and spatially associated with the reworked ancient crustal block (Hou et al. 2015). Sr–Nd isotope data indicate that those

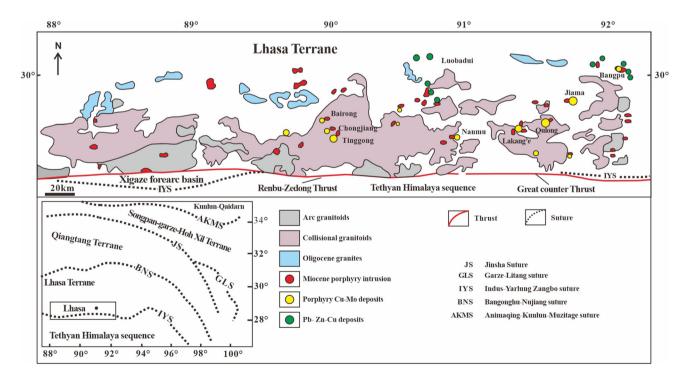


Fig. 1 Tectonic setting and geological map of the Gangdese Porphyry Copper Belt. Modified from Li et al. (2017b)

porphyry Mo deposits are possibly formed by the melting of an ancient lower crust with minor mantle component contribution (Hou et al. 2011, 2015).

3 Deposit geology

The outcrop layer of Jurassic Yeba Formation is widely distributed at the district of Qulong deposit, characterized by intermediate–acid volcanic rock, pyroclastic rock, and a few sedimentary rocks interbedded (Yang et al. 2009; Qin et al. 2014). The intrusive rocks at Qulong deposit are mainly the mid-Miocene intermediate–acid complex (Fig. 2), including the Rongmucuola pluton, the P porphyry, and the X porphyry. The mid-Miocene

Rongmucuola pluton intruded into the Yeba Formation and the Jurassic dacite-rhyolite porphyry at 17.142 ± 0.014 Ma (Li et al. 2017b), which hosts the majority of the Cu-Mo in the Oulong deposit (Li et al. 2017a). It varies gradually from granodiorite to biotite monzogranite in composition (Yang et al. 2009; Zhao et al. 2016), containing plagioclase, K-feldspar, quartz, amphibole, and biotite in variable percentages (Zhao et al. 2016; Li et al. 2017b). The P porphyry (monzogranite porphyry), which was emplaced into the center of the western Rongmucuola pluton at 16.009 \pm 0.016 Ma (Li et al. 2017b), is the syn-ore porphyry at the Oulong deposit (Yang et al. 2009; Hu et al. 2015; Zhao et al. 2016) and is composed of plagioclase (15 vol%), quartz (5 vol%), K-feldspar (5 vol%) and biotite (3 vol%) as phenocrysts, with the

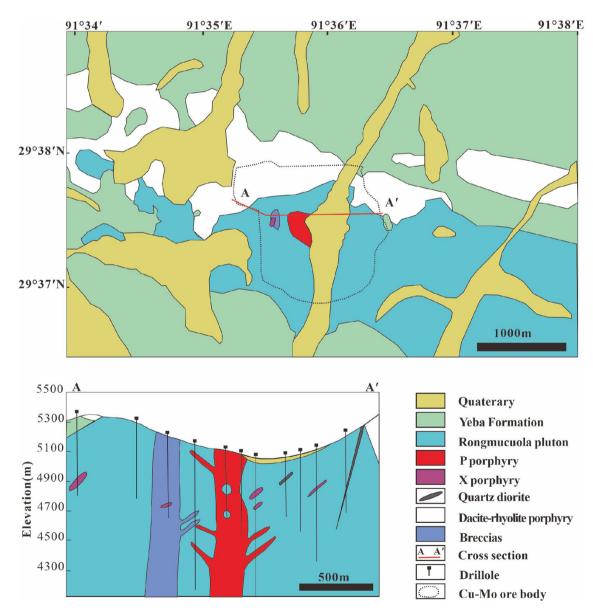


Fig. 2 Geological map a of the Qulong Cu-Mo porphyry deposit with cross-sections, b. Simplified and revised from Zhao et al. (2016)

quartz and feldspar, dominated the groundmass (Yang et al. 2009; Li et al. 2017a). The X porphyry (monzogranite porphyry), which crosscut both the western Rongmucuola pluton and the P porphyry, share a similar composition and texture with the P porphyry except that it contains less biotite.

The mineralization of Qulong PCD is mainly hosted in the western Rongmucuola pluton and associated with the potassic alteration minerals assemblages. The Re-Os age of molybdenite is 16.41 ± 0.48 Ma (Meng et al. 2003), and recently Li et al. (2017b) reported new high precision Re-Os age of 16.126 ± 0.008 Ma and 15.860 ± 0.010 Ma. The ore minerals at the Qulong deposit are dominated by pyrite, chalcopyrite and molybdenite (Fig. 3a, b, c) with minor bornite, galena and sphalerite (Fig. 3d), the mineralization occurs as veins (Fig. 3a, b) or disseminated. Most of the chalcopyrite and molybdenite are presented in veins that consist of quartz, pyrite and minor anhydrite (Fig. 3b). Chalcopyrite is either of disseminated structure that associated with biotite alteration, or occurs as veinlets with an alteration halo of chlorite-sericite. Molybdenite generally filled in the central or margin of quartz veins.

Fluid inclusion studies and H, O isotopic composition of Qulong deposit indicate that the ore–forming fluid of the early stage mineralization and alteration are exsolved from magma at the temperature of 550–650 °C with moderate salinity (9% NaCl), and mixed with meteoric water later (Li et al. 2018; Yang and Hou 2009). The S and Pb isotopes of copper–bearing porphyries and ore minerals from the

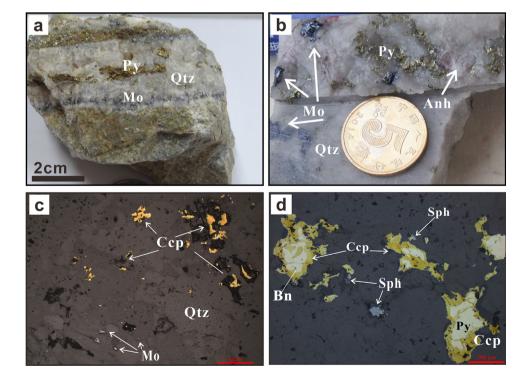
deposit suggest that the ore-forming materials were mainly from magma (Meng et al. 2006).

4 Sampling and analytical methods

The samples used in this study were collected from drill cores of the Qulong Cu-Mo deposit (Fig. 3). Pure pyrite and molybdenite were hand-picked from the crushed ore under a binocular microscope. The measurements were carried out using a GV 5400 mass spectrometer at the Institute of Geochemistry, Chinese Academy of Sciences (IGCAS), Guiyang, China. The sensitivities of the GV5400 for He and Ar were 3.9725×10^{-4} A/Torr and 1.1018×10^{-3} A/Torr, respectively. Pure sulfide grains (1-2 g with sizes of 0.5-1.5 mm) were ultrasonically cleaned in alcohol before being loaded online into vacuum crusher buckets. The samples were baked at 120-150 °C in the crusher buckets for ca. 24 h to remove adhered atmospheric gases. The gases were released by sequential crushing of the pyrite grains in high-vacuum conditions. He and Ar isotopic compositions and abundances were calibrated against pipettes of 0.1 cm³ STP air (5.2 \times 10⁻⁷ cm³ STP ⁴He and 9.3×10^{-4} cm³ STP ⁴⁰Ar). Procedural blanks were $< 2 \times 10^{-10}$ cm³ STP ⁴He and (2–4) $\times 10^{-10}$ cm³ STP ⁴⁰Ar. The detailed analytical procedures are described in Hu et al. (2012).

The sulfur isotope composition of the sulfide minerals was determined using a Thermo MAT253 continuous flow isotope ratio mass spectrometer coupled to an elemental

Fig. 3 Photographs of hand specimens and micrographs from the Qulong PCD. a Sample of Qtz-Py-Mo vein; b sample of Qtz-Py-Mo-Anh vein; c micrograph of Qtz- Mo-Ccp assemblage; d micrograph of Ccp-Bn-Sph-Py assemblage. Abbreviation: Anh-anhydrite, Bn-bonite, Ccp-chalcopyrite, Mo-molybdenite, Py-pyrite, Sph-sphalerite, Qtz-quartz



analyzer (EA–IRMS) at the IGCAS. Measurements are reported using standard δ –notation relative to V–CDT international standard. The reproducibility of replicate analyses of the IAEA international standards: IAEA S1 (- 0.3 ‰), IAEA S2 (+22.62 ‰), and IAEA S3 (- 32.49 ‰) yields a precision that is better than 0.2 ‰ (1 σ).

5 Results

The He and Ar isotopic compositions released from the fluid inclusion of pyrite crystals are listed in Table 1 and Figs. 4, 5, and 6. The concentrations of ³He vary from 4.68 to 14.17×10^{-15} cm³ STP/g, and ⁴⁰Ar concentration range from 8.79 to 154.72×10^{-9} cm³ STP/g. The ³He/⁴He ratios vary from 0.54 to 1.02 Ra (Ra: the ³He/⁴He ratio of air, 1 Ra = 1.39×10^{-6}), and the ⁴⁰Ar/³⁶Ar ratio varies from 300 to 359. δ^{34} S range from - 0.52 to 0.31 ‰, with an average of - 0.12 ‰.

6 Discussion

6.1 Source of ore-forming fluids using He, Ar, and S isotopes

Many studies have demonstrated that pyrite is capable of trap and preserve noble gases (Stuart et al. 1995; Hu et al. 1998, 2004, 2009, 2012; Burnard et al. 1999), and helium and argon trapped in fluid inclusion will not lose extensively within 100 Ma (Burnard et al. 1999). The ratios of ${}^{3}\text{He}/{}^{4}\text{He}$ and ${}^{40}\text{Ar}/{}^{36}\text{Ar}$ will not change even if the trapped He and Ar lost partially (Hu et al. 1997, 2004, 2012; Ballentine and Burnard 2002). Since pyrite is not Libearing mineral, there is no helium produced by the nuclear decay of lithium. Contamination of cosmogenic ³He can also be eliminated since all of the pyrites of this study are collected from drill core of which the depths are more than 50 m. U and Th concentrations in hydrothermal fluids are commonly low (Hu et al. 2012; Norman and Musgrave 1994), therefore the radiogenic ⁴He produced in situ in fluid inclusions is also negligible (Hu et al. 2012). The impact of ⁴⁰Ar produced by the decay of ⁴⁰K in the mineral lattice should be negligible because of the low diffusivity of Ar in pyrite (York et al. 1982; Smith et al. 2001) and the extremely low concentration of K in pyrite (York et al. 1982). The radiogenic ⁴⁰Ar produced in pyrite fluid inclusions is also negligible, due to the high proportion (more than 80%, see the Discussion in 5.1.2) of atmospheric Ar. Therefore, the measured He and Ar isotopic compositions of pyrite samples could basically represent the features of the ore-forming fluids of Qulong porphyry Cu-Mo deposit.

6.1.1 Helium

The ³He/⁴He ratios of pyrites from the Qulong PCD are between 0.54 and 1.01 Ra (Table 1; Fig. 5). By calculating the F^4 He values, which is defined as 4 He/ 36 Ar value of the sample relative to the atmospheric ⁴He/³⁶Ar value of 0.1655 (Kendrick et al. 2001), we can determine the atmospheric Helium contribution. The F⁴He values of Oulong PCD are between 300 and 1920 (Table 1), indicating that ⁴He of the Qulong PCD is 300-1900 times higher than the atmospheric concentration. This means that the atmospheric He contribution is negligible in the mineralizing fluids. The ³He/⁴He ratios of the Qulong PCD are significantly higher than the crustal fluid value (0.01–0.05Ra) and lower than the mantle components value (6-9 Ra), indicating that both mantle and crustal components were involved in the hydrothermal fluids of the Qulong PCD. The ³He/⁴He and ⁴⁰Ar/³⁶Ar ratios of Oulong PCD is plotted on the trend line (Fig. 5) modeled by binary mixing between a high ${}^{3}\text{He}/{}^{4}\text{He}$ (1.23 Ra), high ⁴⁰Ar/³⁶Ar fluid containing mantle component, and a crustal fluid with atmospheric ⁴⁰Ar/³⁶Ar and radiogenic ⁴He (Burnard et al. 1999). The Qulong porphyry deposit is spatially and temporally associated with monzogranite porphyry (P porphyry) (Meng et al. 2003; Qin et al. 2014). S, Pb, H, and O isotopic features of the copper-bearing porphyries and ore minerals from the Qulong deposit show that ore-forming fluid and materials were mainly derived from the magma (Meng et al. 2006; Yang and Hou 2009). Therefore, the high ${}^{3}\text{He}/{}^{4}\text{He}$ ratio mantle end-member in the ore-forming fluid was likely to be exsolved from the monzogranite porphyry (P porphyry) magma. The ³He/⁴He ratio (1.02 Ra, the maximum value measured in this study) of the mantle end-member is much less than the typical sub-continental mantle value (~ 6 Ra), indicating that the mantle helium from mantle-derived magma was diluted by incorporating radiogenic helium within the parental magma of the associated porphyry before helium was released into the hydrothermal system (Hu et al. 2012). This is consistent with the conclusion that monzogranite porphyry magma was generated by partial melting of the thickened lowercrust under the Tibetan orogen with the input of mantle components (Meng et al. 2006; Yang et al. 2015).

We can use the 3 He/ 4 He ratios to calculate the proportion of mantle and crustal components based on binary mixing. The proportion of mantle or crustal 4 He can be calculated using the following equation (Burnard et al. 1999):

$${}^{4}\text{He}_{\text{total}} = {}^{4}\text{He}_{\text{mantle}} + {}^{4}\text{He}_{\text{crust}}$$
$$= {}^{4}\text{He}_{\text{crust}} + {}^{3}\text{He}_{\text{mantle}} \times \left({}^{4}\text{He} / {}^{3}\text{He} \right)_{\text{mantle}}$$

Table 1 H	le, Ar and S iso	stopic compo	ositions of sulfide:	Table 1 He, Ar and S isotopic compositions of sulfides from the Qulong porphyry deposit	orphyry deposi	it						
Sample	Mineral	Weight (g)	$^{40}_{ m Ar}$ (10 ⁻⁸ cm ³ STP)	${}^{36}{ m Ar}_{(10^{-10})}_{ m cm^3 STP)}$	$^{40}\mathrm{Ar}/^{36}\mathrm{Ar}$	4 He $(10^{-8}$ cm 3 STP)	${}^{3}_{\rm He}$ (10 ⁻¹⁴ cm ³ STP)	³ He/ ⁴ He (Ra)	³⁶ Ar/ ³ He	F ⁴ He	${}^{36}\mathrm{Ar}{}^{3}\mathrm{He} F^{4}\mathrm{He} {}^{40}\mathrm{Ar}{}^{*}{}^{4}\mathrm{He} {}^{334}\mathrm{S} (\%)$	δ ³⁴ S (‰)
zk816-1	Pyrite	4.16	18.6 ± 0.07	5.16 ± 0.07	359 ± 5	2.60 ± 0.08	1.95 ± 0.12	0.54 ± 0.04	26515	303	1271	-0.52
q11513	Pyrite	3.11	9.2 ± 0.04	2.82 ± 0.05	326 ± 5	2.62 ± 0.08	3.04 ± 0.14	0.83 ± 0.05	9283	561	325	0.24
zk1616-1	Pyrite	2.52	34.8 ± 0.14	10.4 ± 0.01	334 ± 3	2.18 ± 0.07	3.08 ± 0.14	1.02 ± 0.06	33,757	126	1843	-0.05
zk816-2	Pyrite	1.94	1.7 ± 0.01	0.57 ± 0.29	300 ± 15	1.81 ± 0.05	2.52 ± 0.12	1.01 ± 0.06	2250	1922	15.1	-0.19
q11518	Pyrite	1.59	24.6 ± 0.10	8.18 ± 0.10	300 ± 3	1.77 ± 0.05	2.25 ± 0.12	0.91 ± 0.06	36,369	130	214	-0.25
zk816-3	Pyrite											-0.32
zk816-3	Molybdnite											- 0.48
QLD2	Pyrite											0.31
QLD2	Molybdnite											0.21
Errors quot ⁴⁰ Ar*is the	Errors quoted are at the 1 σ confidence level ⁴⁰ Ar*is the non-atmospheric Ar, ⁴⁰ Ar* = ⁴⁰ ,	σ confidenc ric Ar, ⁴⁰ Ar	Errors quoted are at the 1 σ confidence level 40 Ar*is the non-atmospheric Ar, 40 Ar* = 40 Ar-[36 Ar × 295.5]	295.5]								

 $F^{4}He{=}(^{4}He)^{36}Ar)_{sample}/(^{4}He)^{36}Ar)_{air},\ (^{4}He)^{36}Ar)_{air}{=}0.1655$

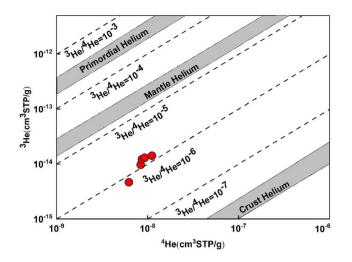


Fig. 4 Helium isotopes data of volatiles released from fluid inclusion of pyrite in the Qulong porphyry Cu–Mo deposit (modified from Mamyrin and Tolstikhin 1984)

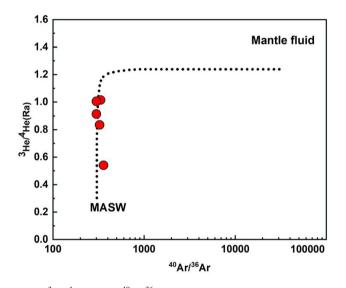


Fig. 5 ${}^{3}\text{He}/{}^{4}\text{He}$ versus ${}^{40}\text{Ar}/{}^{36}\text{Ar}$ for pyrite from the Qulong Cu–Mo deposit. He and Ar isotopic compositions of Qulong Cu–Mo deposit plotted on the trend line modelled by binary mixing between a "magmatic fluid" with ${}^{3}\text{He}/{}^{4}\text{He} = 1.23 \text{ Ra}, {}^{40}\text{Ar}/{}^{36}\text{Ar} \ge 40,000$ and a crust fluid with ${}^{40}\text{Ar}/{}^{36}\text{Ar} = 295.5, {}^{3}\text{He}/{}^{4}\text{He} = 0.01 \text{ Ra}.$ Modified from Burnard et al. (1999)

The total ⁴He is measured from the samples; since ³He are mainly from mantle so the measured ³He could be regarded as mantle ³He roughly. The mantle ⁴He/³He is about 1.19×10^5 (Burnard et al.1999). According to the calculation, around 8%–90% ⁴He was derived from the crust, corresponding to a mantle contribution of around 10%–16%.

6.1.2 Argon

The measured 40 Ar/ 36 Ar of Qulong deposit range from 300 to 359, which are higher than the value of air-saturated

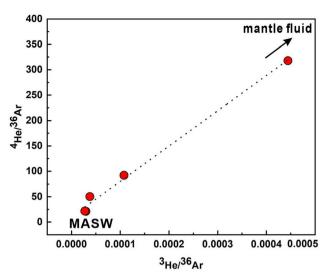


Fig. 6 3 He/ 36 Ar versus 4 He/ 36 Ar for pyrite from the Qulong Cu–Mo deposit. The linear correlation is obtained by least squares regression: 4 He/ 36 Ar = 10.7 + 694,978 × 3 He/ 36 He, R^{2} = 0.99

water (ASW, 40 Ar/ 36 Ar = 295.5) and indicate the presence of excess 40 Ar* derived from the mantle or crustal components. We can calculate the proportion of 40 Ar* with the following equation (Kendrick et al. 2001):

$${}^{40}\text{Ar} * \% = \frac{\left({}^{40}\text{Ar}/{}^{36}\text{Ar}\right)\text{sample} - 295.5}{\left({}^{40}\text{Ar}/{}^{36}\text{Ar}\right)\text{sample}} \times 100$$

Estimated 40 Ar* range from 1.6% to 17%, with an average of 8.4%, which means that the proportion of air derived Ar is 83%-98.3%, with an average of 91.6%. Therefore, the ⁴⁰Ar of mineralization fluid of Qulong is dominated by atmospheric Ar that is commonly provided by the meteoric water. Since both ³He and ³⁶Ar are unradiogenic, the ³He/³⁶Ar ratio of meteoric water before mixing with mantle fluid is a constant value of 5×10^{-8} (Burnard et al. 1999). By extrapolating the trend in Fig. 6 to pure ASW (³⁻ $\text{He}/^{36}\text{Ar} = 5 \times 10^{-8}$) generate a ${}^{4}\text{He}/{}^{36}\text{Ar}$ value of 10.7, which is much higher than that of ASW (0.46, Ozima and podosek 2002), indicating that the ASW was modified by the addition of radiogenic ⁴He. Therefore, a crustal fluid end member consists of ASW modified by the crustal radiogenic ⁴He, called modified air-saturated water (MASW), is considerable. Two samples have high 40 Ar*/⁴He ratios (Fig. 7) that much higher than that of the mantle and crust production ratios (0.5 and 0.2 respectively). The higher ${}^{40}\text{Ar}*/{}^{4}\text{He}$ ratios may cause by the addition of a third source for ⁴⁰Ar* or loss of He. The measured ⁴He values are an order of magnitude lower than other deposit in this area (such as Bangpu deposit, Wang et al. 2015), therefore the higher ⁴⁰Ar*/⁴He ratios may because of boiling of the ore-forming fluid, which causes the enrichment of less volatile heavier noble gases in the

residue fluid and the lighter and more volatile noble gases enriched in the vapor phase (Kendrick et al. 2001). This is consistent with the fluid inclusion studies that boiling occurred in the early stage fluid (Yang and Hou 2009).

6.1.3 Sulfur

The δ^{34} S values of sulfides from the Qulong Cu–Mo deposit have a narrow range (- 0.52 ‰–0.31 ‰) and are within the range of the magmatic value (0 ‰ ± 2 ‰; Kyser 1990), indicating a magmatic origin. This may due to the low concentration in the MASW compared with the magmatic hydrothermal fluid. In addition, the country rocks of the deposit are mainly volcanic rocks so that the contamination of sulfur from these rocks would not change the S isotopic compositions of magma.

6.2 Comparison between porphyry Cu and Mo deposits in the Gangdese belt

Bangpu is a Mo-dominant with Cu as a by-product deposit in Qulong–Jiama ore–district in Gangdese metallogenic belt. The 3 He/ 4 He ratios of Bangpu (0.12–0.36 Ra) are lower than that of Qulong deposit (0.54–1.02 Ra), and the estimated proportion of crustal derived He for the Bangpu deposit and the Qulong deposit is 94%–98% and 84%– 90%, respectively, which means Qulong deposit contains more mantle components compared with the Bangpu deposit, which is consistent with the results of Pb isotopes (Yao et al. 2002; She et al. 2005; Li et al. 2012; Zhou et al. 2012; Meng et al. 2006). This is coincident with the fact that the mineralization of the Qulong deposit is due to the refertilization of Cu in the juvenile lower crust with

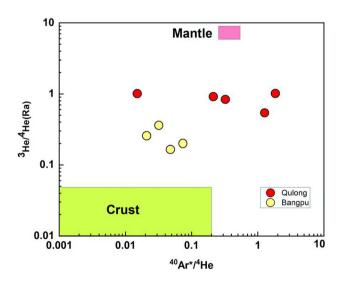


Fig. 7 40 Ar*/⁴He versus 3 He/⁴He for sulfides from the Qulong Cu– Mo and Bangpu Mo–Cu deposits in Gangdese metallogenic belt. Data of Bangpu porphyry Mo deposit collected from Wang et al.(2015)

significant mantle components, whereas the Bangpu porphyry Mo deposit is related to remelting of an ancient lower crust with minor mantle component contributions (Hou et al. 2013, 2015).

7 Conclusion

The ore-forming fluids of the Qulong Cu–Mo deposit were a mixture of a fluid containing mantle components and a crustal fluid containing atmospheric Ar and crustal radiogenic ⁴He. The former was probably exsolved from the orebearing porphyry magma, and the later was meteoric water interacted with crustal rocks, therefore modified by the addition of radiogenic ⁴He. δ^{34} S values of sulfides from Qulong deposit indicate that the sulfur is of magmatic origin and crustal fluid does not supply significant S to the ore-forming system. Mantle-derived components played an important role in the formation of Miocene deposits in the Gangdese metallogenic belt. Cu–Mo deposit contains much more mantle components compared with the Mo–Cu deposit.

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Compliance with ethical standards

Conflict of interest The authors declare that there is no conflict of interest.

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